## QUANTUM MECHANICS I

## **PHYS 516**

## Problem Set # 1 Distributed: Jan. 8, 2010 Due: Jan. 20, 2010

The Frobenius Method is a classical workhorse for finding solutions of relatively simple ordinary differential equations. (The link is to the course website: http://www.physics.drexel.edu/~bob/PHYS516\_10/Frobenius.pdf). Here are two examples:

$$\left(\frac{d^2}{dr^2} + \frac{A}{r^2} + \frac{B}{r} + C\right) R(r) = 0$$
$$\left(\frac{d^2}{dr^2} + \frac{D}{r^2} + E + Fr^2\right) R(r) = 0$$

The first is a general form for the radial part of the quantum wavefunction for a Coulomb potential. The second is a general form for the radial part of the quantum wavefunction for a harmonic oscillator potential.

1. Use the Frobenius method to determine the quantization condition for square-integrable radial functions  $\int_0^\infty R^2(r)dr < \infty$ :

**Coulomb:** on the coefficients A, B, C;

**Harmonic Oscillator:** on the coefficients D, E, F.

**2.** For the Coulomb problem in three dimensions determine the coefficients A, B, C in the

**a.** relativistic case (Klein-Gordan Equation)

**b.** nonrelativistic case (NR Schrödinger Equation).

**3.** Determine the coefficients D, E, F for the nonrelativistic harmonic oscillator in

 $\mathbf{i.}$  one dimension

ii. two dimensions

iii. three dimensions

4. Compute the energy of an electron (in eV) in the most tightly bound state about each of these nuclei:

|                               | Z = 1  | Z = 26       | Z = 82       |
|-------------------------------|--------|--------------|--------------|
|                               | Proton | Iron Nucleus | Lead Nucleus |
| N.R. Schrödinger Equation     | -13.58 |              |              |
| Relativistic Klein-Gordon Eq. |        |              |              |

In computing the relativistic energy, subtract off the electron rest energy  $mc^2$  and enter  $E - mc^2$  in the table above.

In this problem, if there are any surprises, explain:

## **a.** What?

- **b.** Where?
- c. Why?
- **d.** What does it mean?
- e. What to do about it?