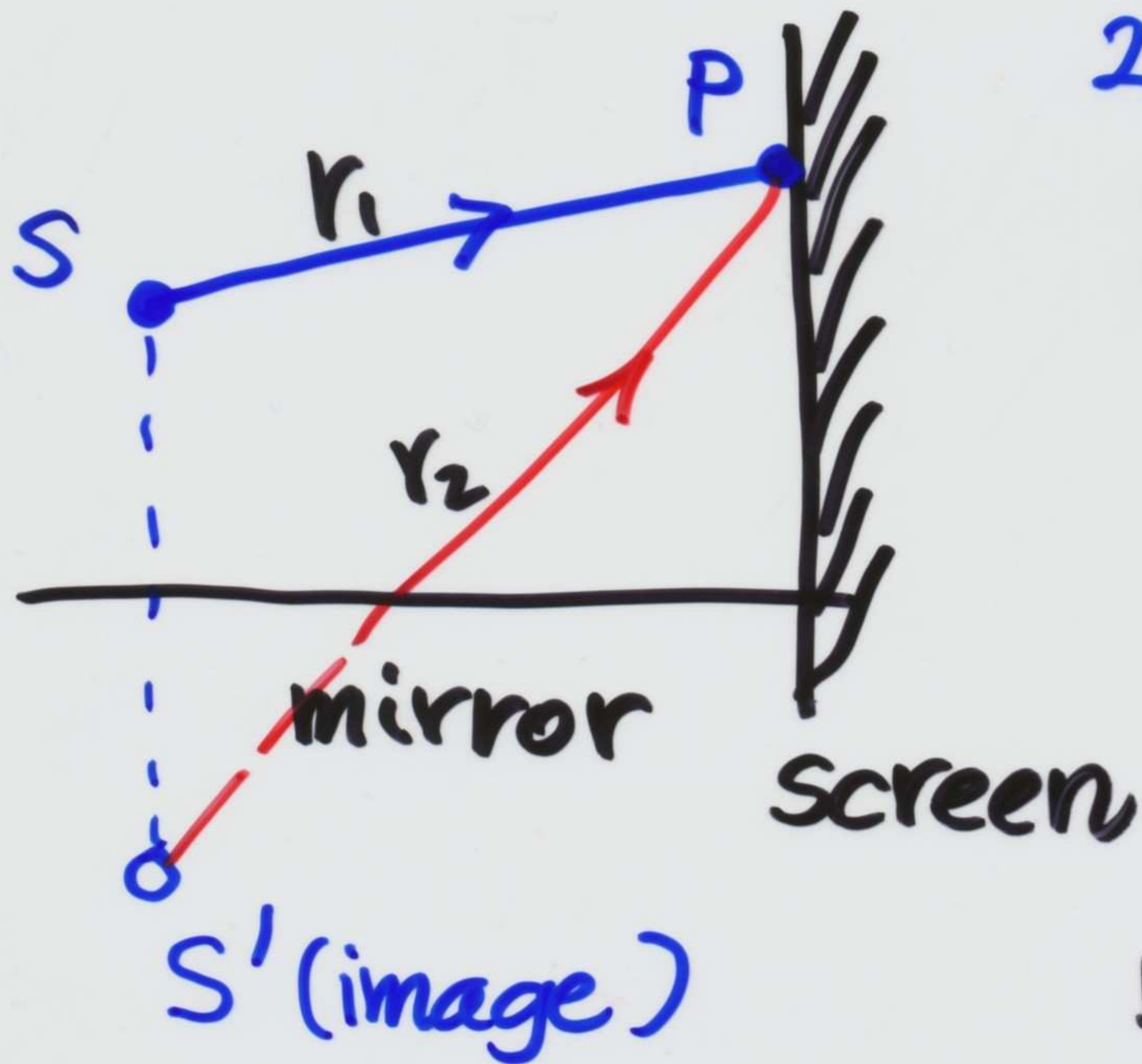


Phase change due to reflection

Lloyd's mirror



2 paths SP & $S'P$

$$\delta = r_2 - r_1 \Rightarrow \text{interference}$$

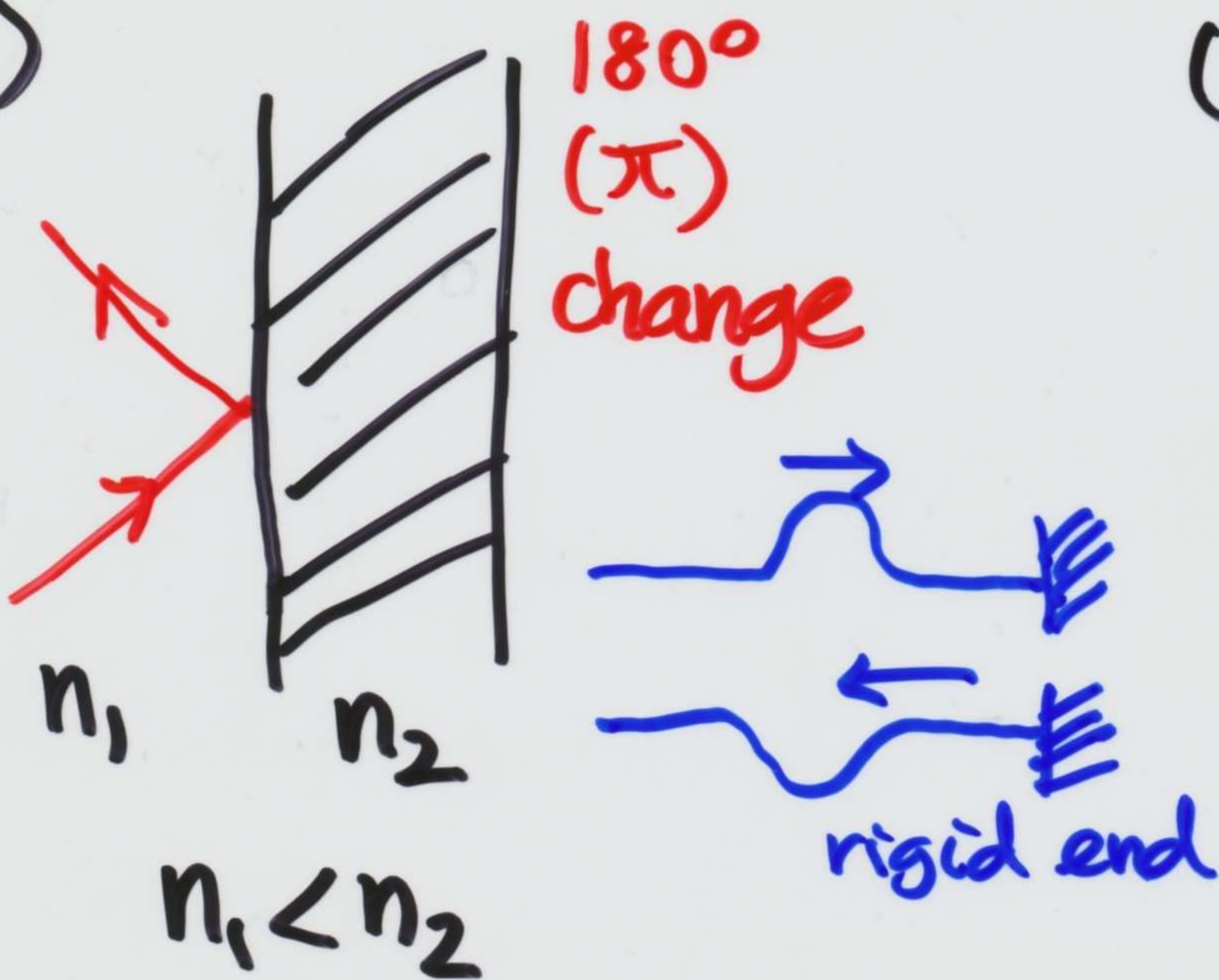
BUT

the positions of bright & dark fringes are

REVERSED

because coherent sources S & S' differ in phase by 180° (π)

(A)

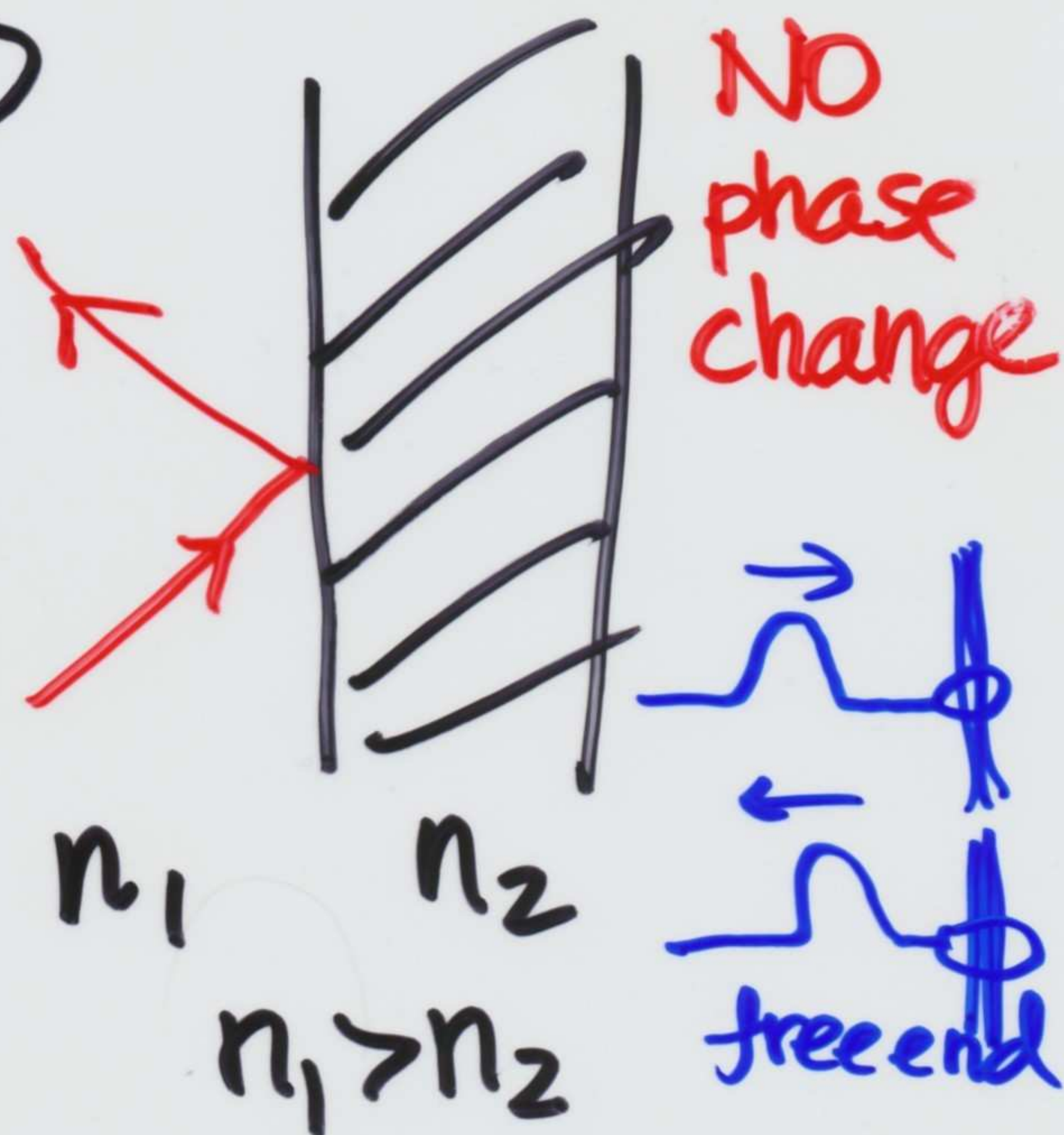


180° (π) change

rigid end

$$n_1 < n_2$$

(B)



NO phase change

free end

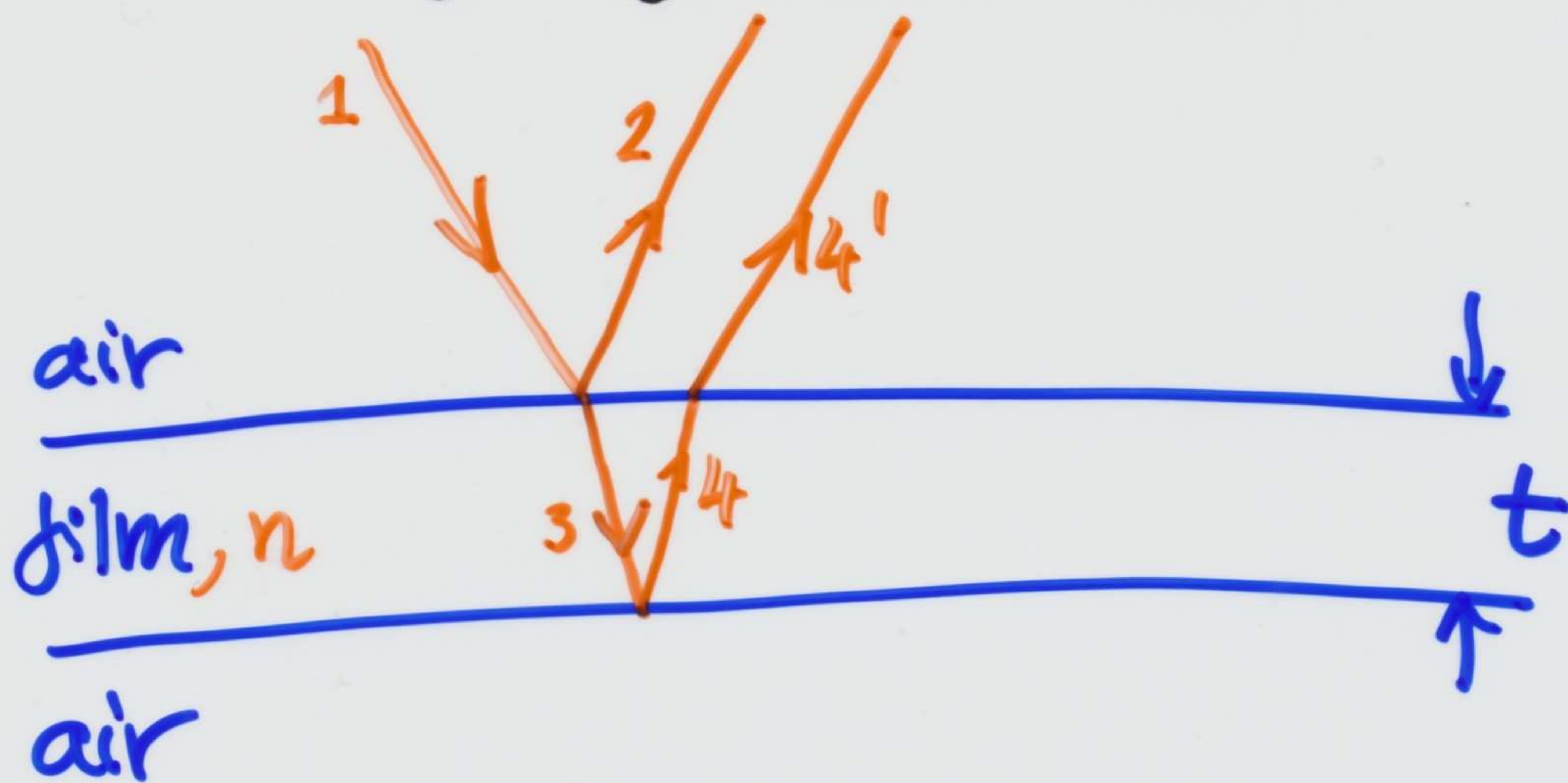
$$n_1 > n_2$$

$n_1, n_2 \dots$ index of refraction

Interference in thin films

- oil film on water
- soap bubble reflections

Consider a film of uniform thickness t & index of refraction n



→ ray 1: reflected ray 2 undergoes a phase change 180° ($n > 1$, for air $n_{\text{air}} = 1$)

→ ray 3: reflected ray 4 (& 4') \Rightarrow no phase change $n_{\text{air}} < n$

→ inside the film $\lambda_n = \frac{\lambda}{n}$
(changed wavelength)

Rule for the constructive interference:

$$2t = (m + \frac{1}{2}) \lambda_n \quad (m = 0, 1, 2, \dots)$$

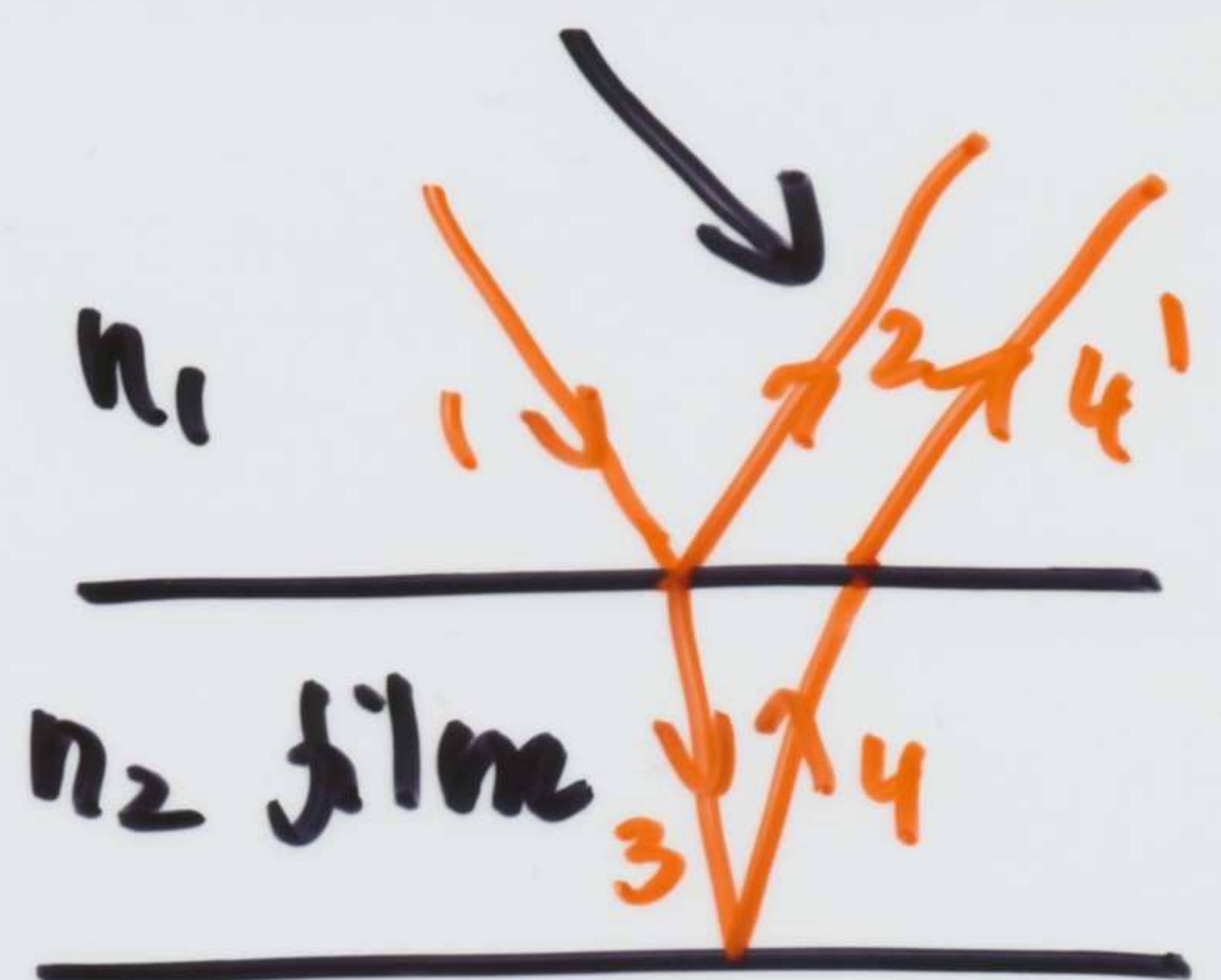
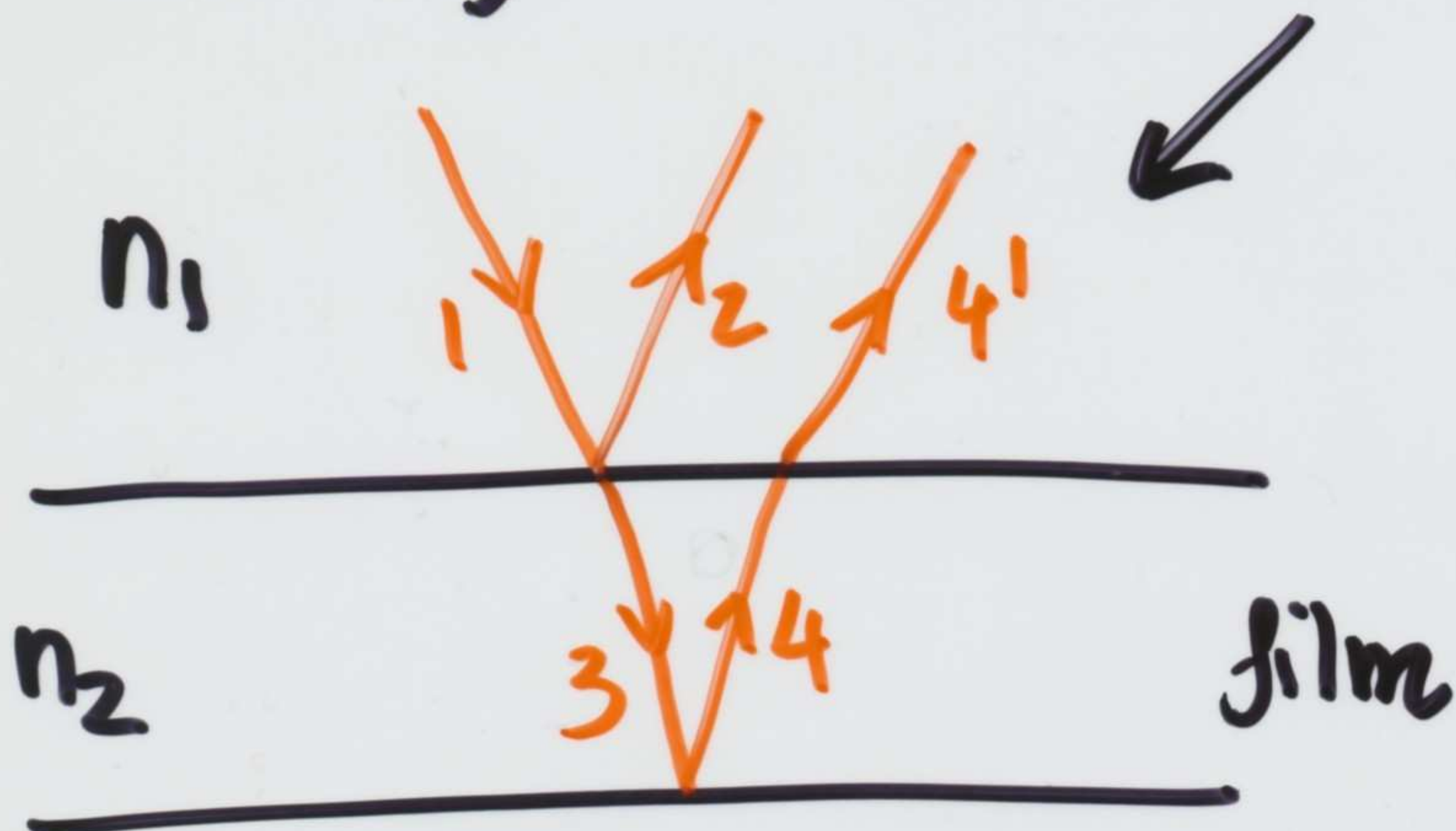
OR

$$2nt = (m + \frac{1}{2}) \lambda \quad (m = 0, 1, 2, 3, \dots)$$

Rule for the destructive interference:

$$2nt = m \lambda \quad (m = 0, 1, 2, 3, \dots)$$

BUT if $n_1 < n_2 < n_3$ or $n_1 > n_2 > n_3$



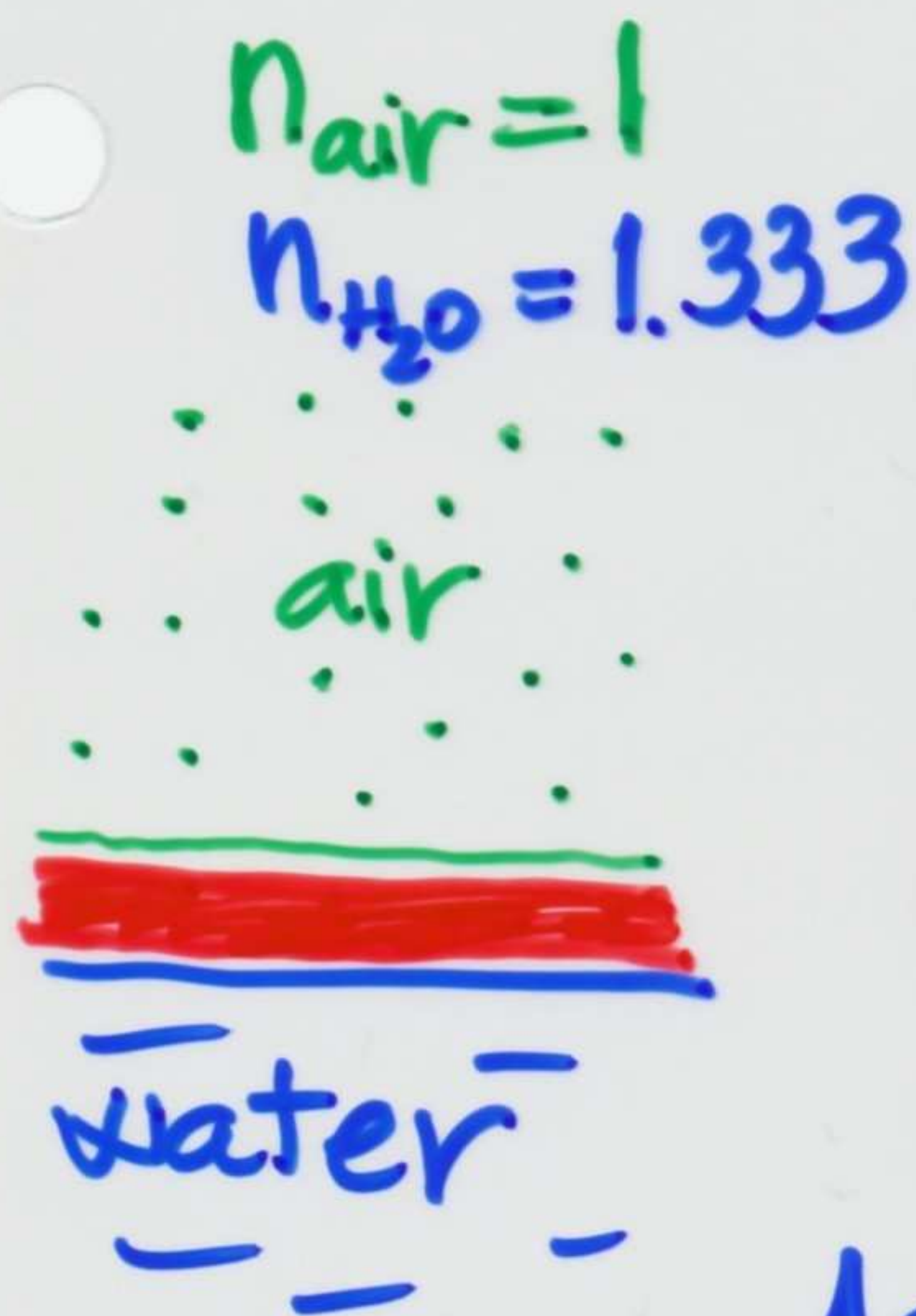
n_1 $1 \rightarrow 2$ 180°
 n_3 $3 \rightarrow 4$ 180°

n_1 $1 \rightarrow 2$ NO
 n_3 $3 \rightarrow 4$ NO

constructive: $2nt = m\lambda$
 destructive: $2nt = (m + \frac{1}{2})\lambda$

REVERSED
 criteria 27-8

Quiz: You spill two liquids onto water. They both form thin films on the water surface but film 1 appears dark & film 2 appears bright in reflected light. Explain!



Assume: $n_{\text{film 1}} > n_{\text{air}}$ & $n_{\text{film 2}} > n_{\text{air}}$

(a) $n_{\text{film 1}} > n_{\text{H}_2\text{O}}$ & $n_{\text{film 2}} > n_{\text{H}_2\text{O}}$

(b) $n_{\text{film 1}} < n_{\text{H}_2\text{O}}$ & $n_{\text{film 2}} < n_{\text{H}_2\text{O}}$

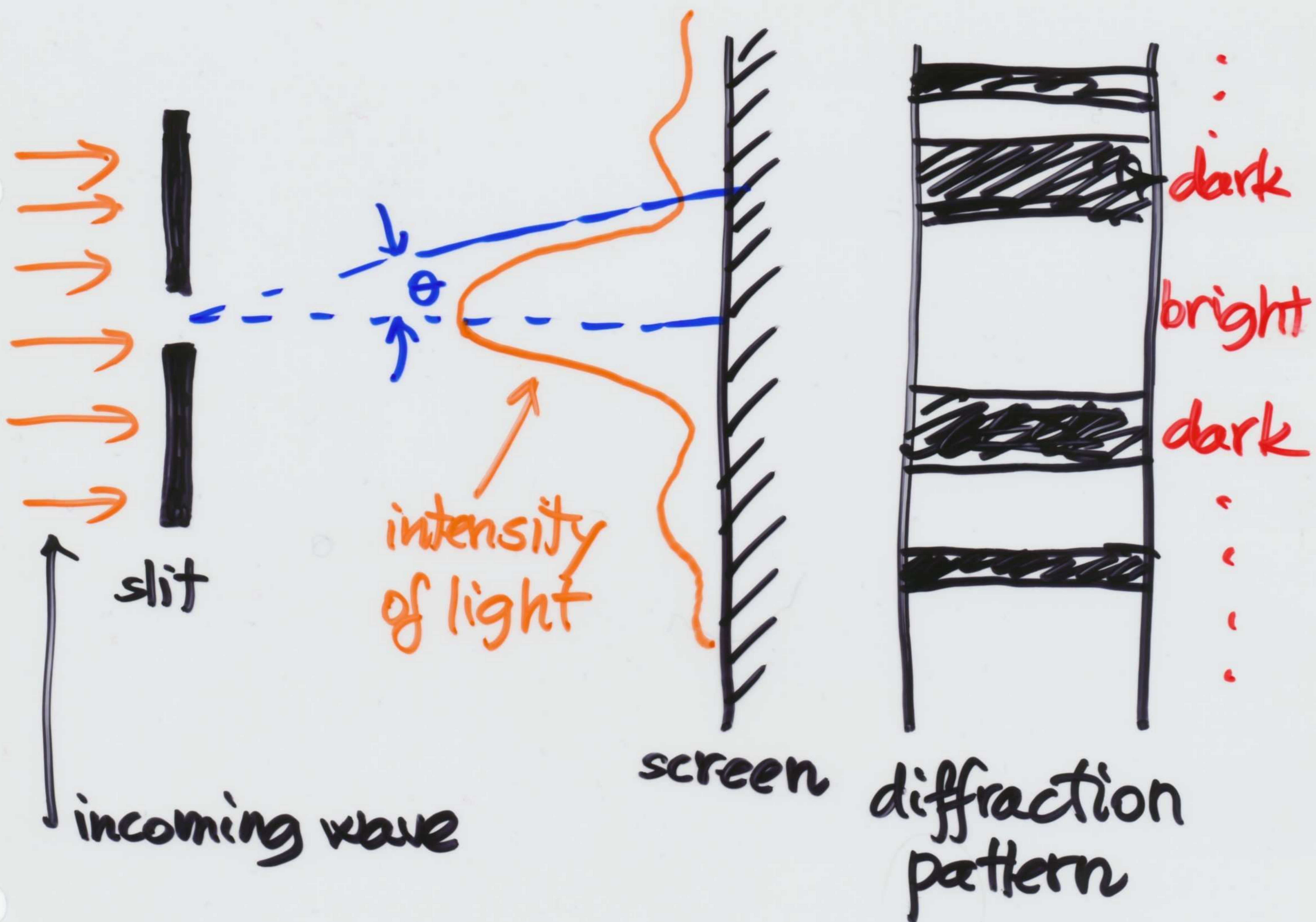
~~(c)~~ $n_{\text{film 1}} < n_{\text{H}_2\text{O}}$ & $n_{\text{film 2}} > n_{\text{H}_2\text{O}}$

~~(d)~~ $n_{\text{film 2}} < n_{\text{H}_2\text{O}}$ & $n_{\text{film 1}} > n_{\text{H}_2\text{O}}$

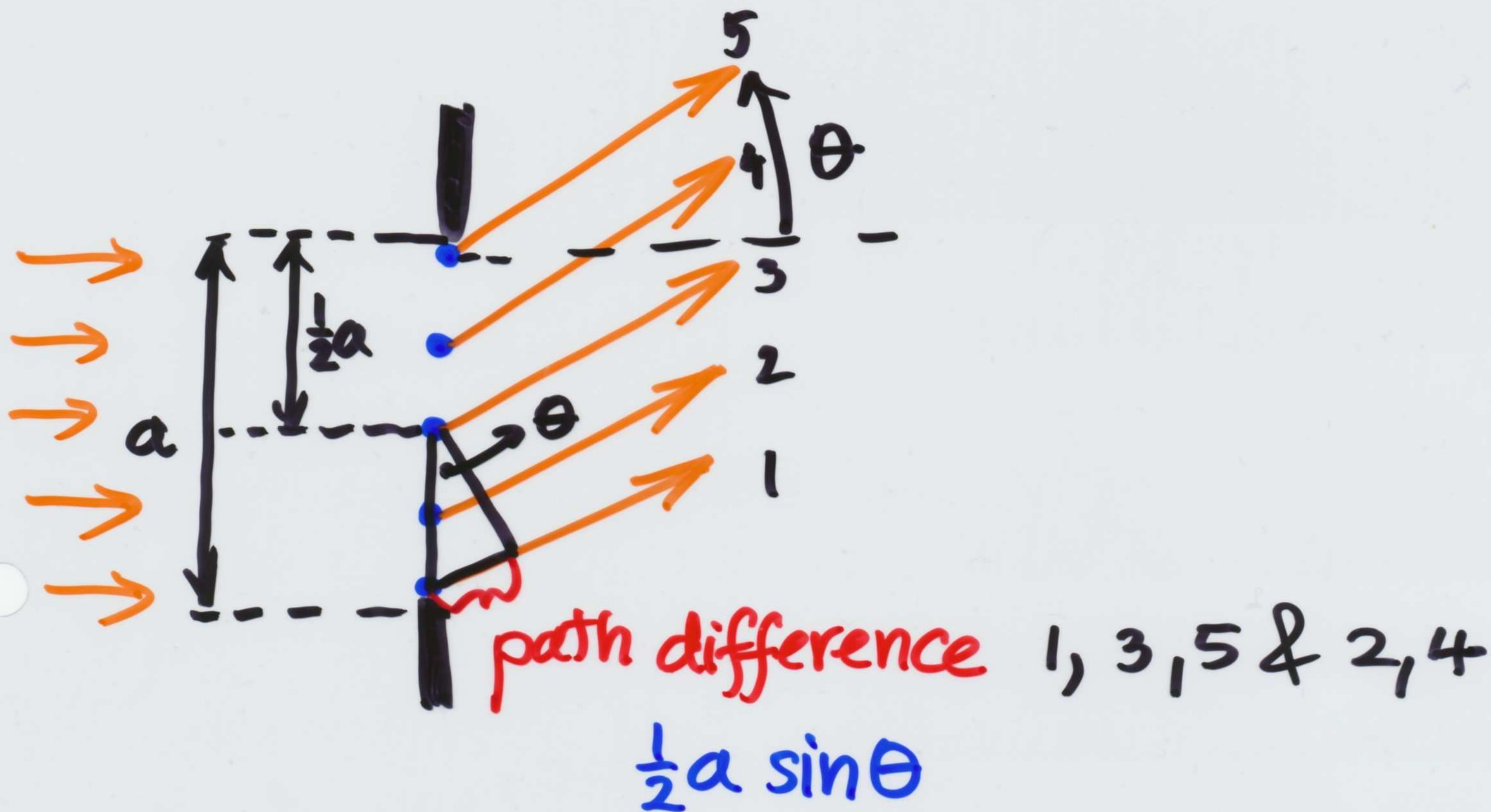
Diffraction Patterns

→ EM waves pass through small openings & around obstacles & by sharp edges

Consider light passing through a narrow opening, slit, & projected onto a screen:



Mathematical model for single-slit Fraunhofer diffraction pattern



Huygens principle: each portion of
the slit acts as a source of waves

⇓
interference

Where is the first dark band?

$$\frac{a}{2} \sin \theta = \frac{\lambda}{2} \quad (\text{phase change } \pi, 180^\circ)$$

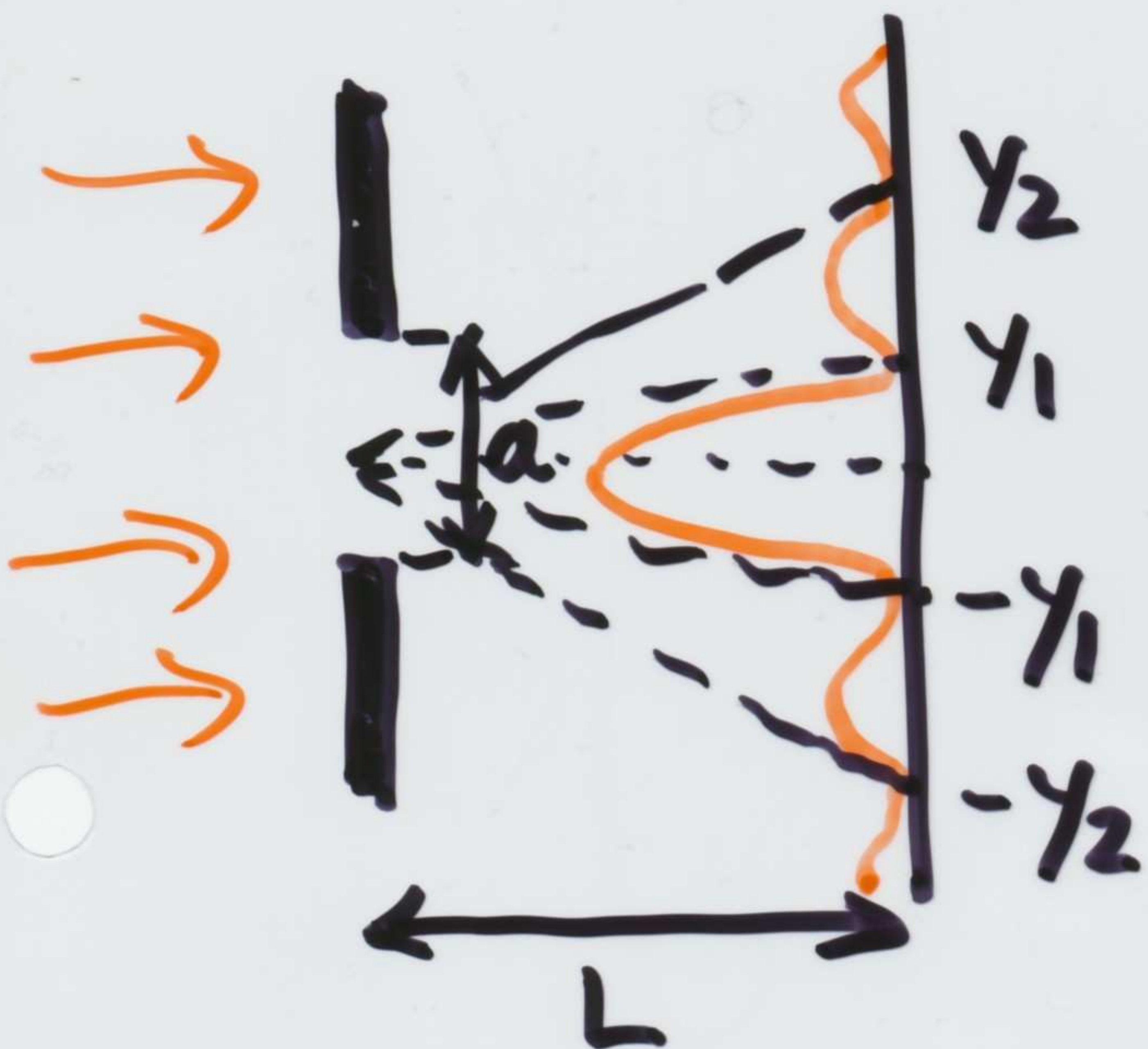
$$\Downarrow$$

$$\sin \theta = \frac{\lambda}{a}$$

General condition for destructive interference

$$\sin \theta_{\text{dark}} = m \frac{\lambda}{a} \quad m = \pm 1, \pm 2, \pm 3, \dots$$

(but NOT $m=0$)



$$\sin \theta_2 = \frac{2\lambda}{a}$$

$$y_2 = L \tan \theta_2$$

$$\sin \theta_1 = \frac{\lambda}{a}$$

$$y_1 = L \tan \theta_1$$

$$\sin \theta_{-1} = -\frac{\lambda}{a}$$

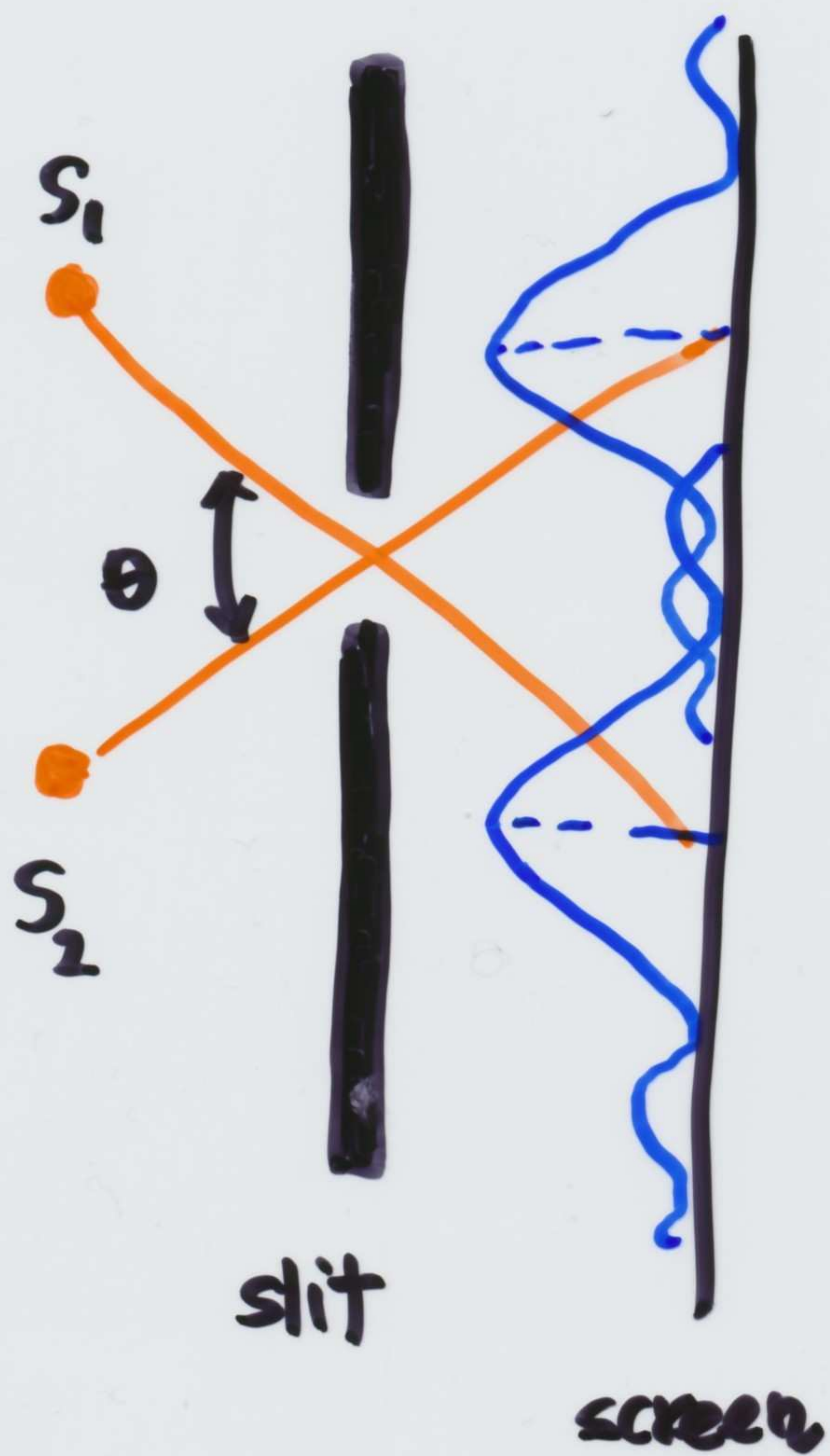
$$y_{-1} = L \tan \theta_{-1}$$

$$\sin \theta_{-2} = -\frac{2\lambda}{a}$$

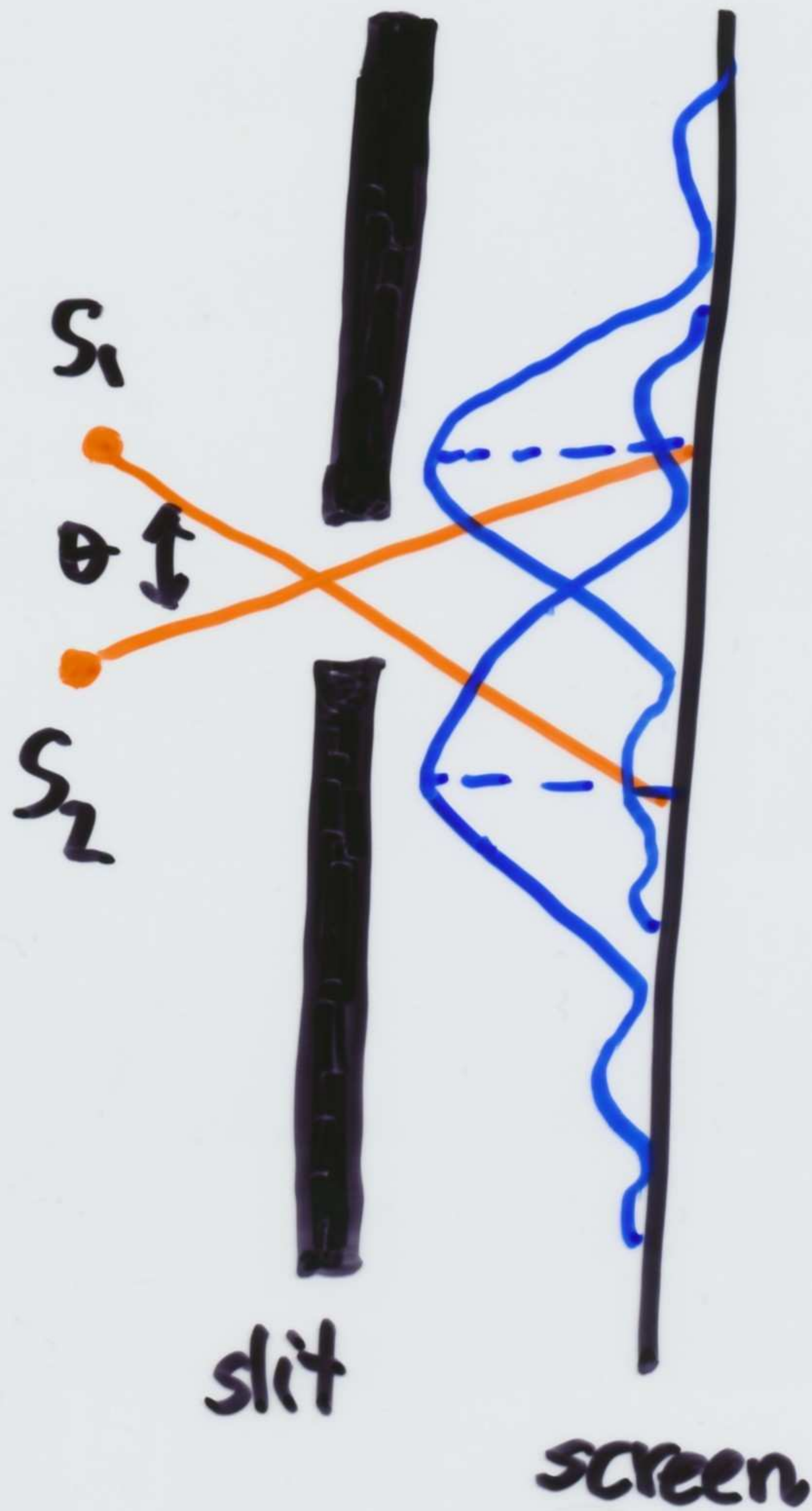
$$y_{-2} = L \tan \theta_{-2}$$

Resolution of Single-Slit & Circular Apertures

When an instrument or your eye can distinguish between two objects (light sources) \Rightarrow the two objects are **RESOLVED**



RESOLVED



**BARELY
RESOLVED**

Rayleigh's criterion: The central maximum of the diffraction pattern of one source falls on the first minimum of the diffraction pattern of the other source



the two sources are just
RESOLVED

Minimum angular separation θ_{\min} :

$$\sin \theta_{\min} = \frac{\lambda}{a}$$

λ ← wavelength of EM wave (light)
 a ← width of a slit

$\lambda \ll a$ (most situations)

⇓ $\sin \theta \approx \theta$

$$\theta_{\min} = \frac{\lambda}{a} \text{ [in radians!]}$$

The angle between 2 sources, S_1 & S_2 , has to be $> \theta_{\min}$:

$\theta > \theta_{\min} \Rightarrow$ RESOLVED.

Circular aperture of diameter D



$$\theta_{\min} = \frac{1.22 \lambda}{D}$$

for circular apertures

For example:

D ... diameter of a pupil

OR

diameter of the astronomical telescope

Quiz: You are observing a binary star (2 stars close by). Which λ of light you should choose to MAXIMIZE resolution?

(a) red (b) yellow (c) green

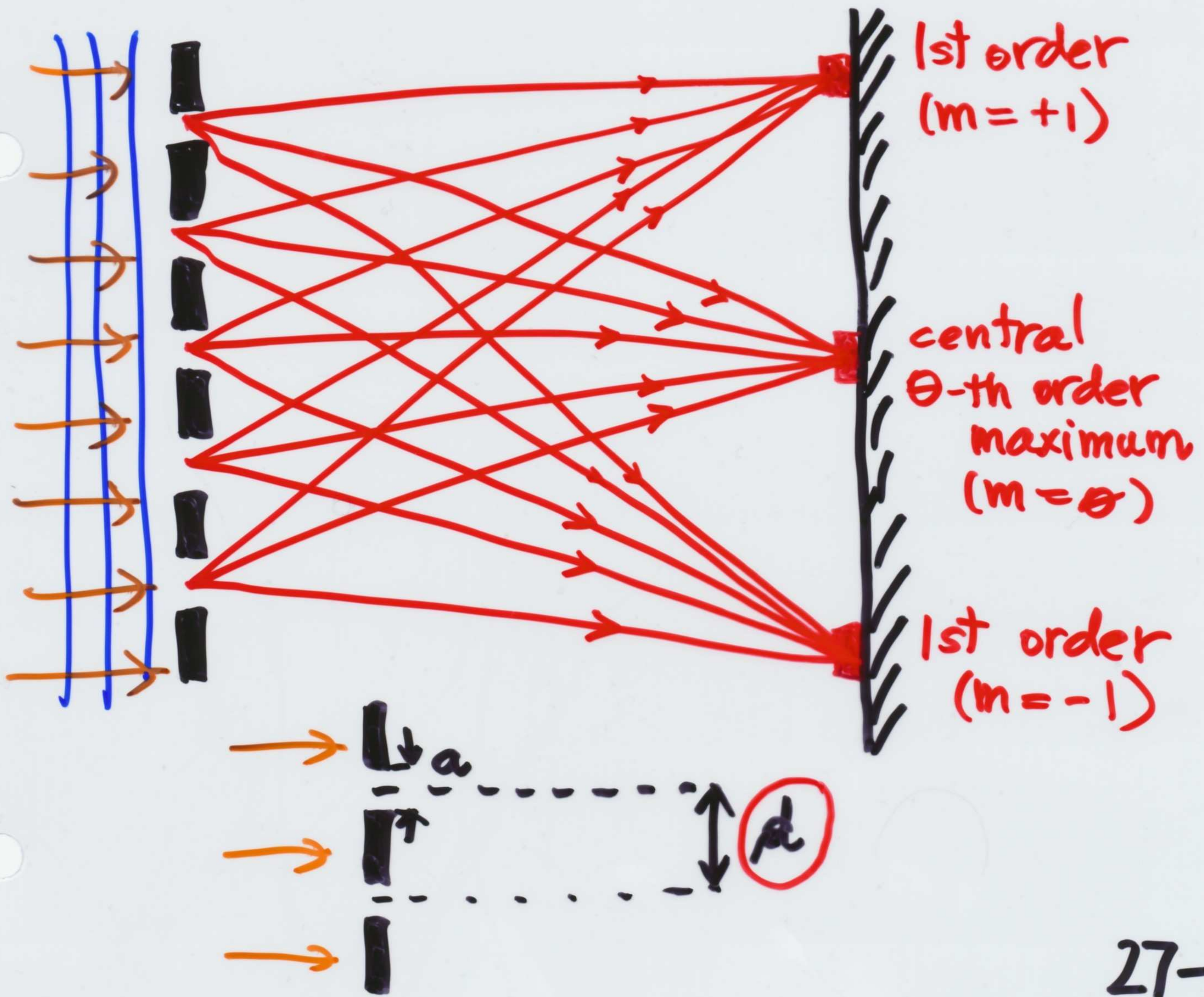
~~(d) blue~~

The Diffraction Grating = device

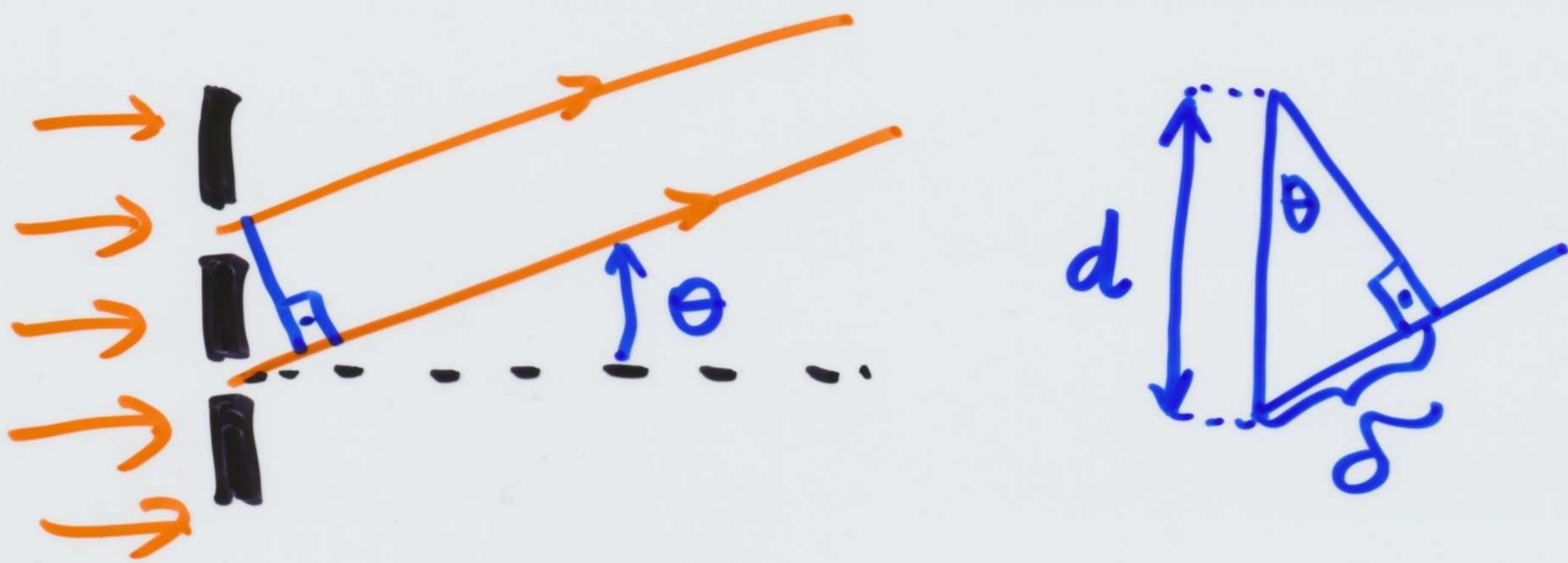
for analysing light sources consists of large # of equally spaced slits, e.g. ~~5000~~ slits per cm

↓

$$\text{slit spacing } d = \frac{10^{-2} \text{ m}}{5000} = 2 \times 10^{-6} \text{ m}$$
$$= \underline{\underline{2 \mu\text{m}}}$$



Warning: here the size of the slit a is not the parameter which determine the maxima of the diffraction \rightarrow interference pattern but rather the spacing d BETWEEN slits



δ ... path difference

$$\delta = d \cdot \sin \theta$$

Condition for constructive interference:

$$d \cdot \sin \theta_{\text{bright}} = m \cdot \lambda$$

$$m = 0, \pm 1, \pm 2, \dots$$

Quiz: Red light of $\lambda = 700\text{nm}$ is used to form interference pattern on a screen a distance L away. The spacing between slits is d . We mark the bright maxima on the screen, then replace the red light source with UV source with $\lambda_1 = \frac{1}{2}\lambda = 350\text{nm}$. If we want to match the marks on the screen with the new bright maxima of λ_1 source, we have to:

(a) move the screen to $2L$

(b) — " — to $\frac{1}{2}L$

(c) double the distance $d_1 = 2d$ between the slits

~~(d) half the distance $d_1 = \frac{1}{2}d$ between the slits~~

Diffraction of X-rays by Crystals

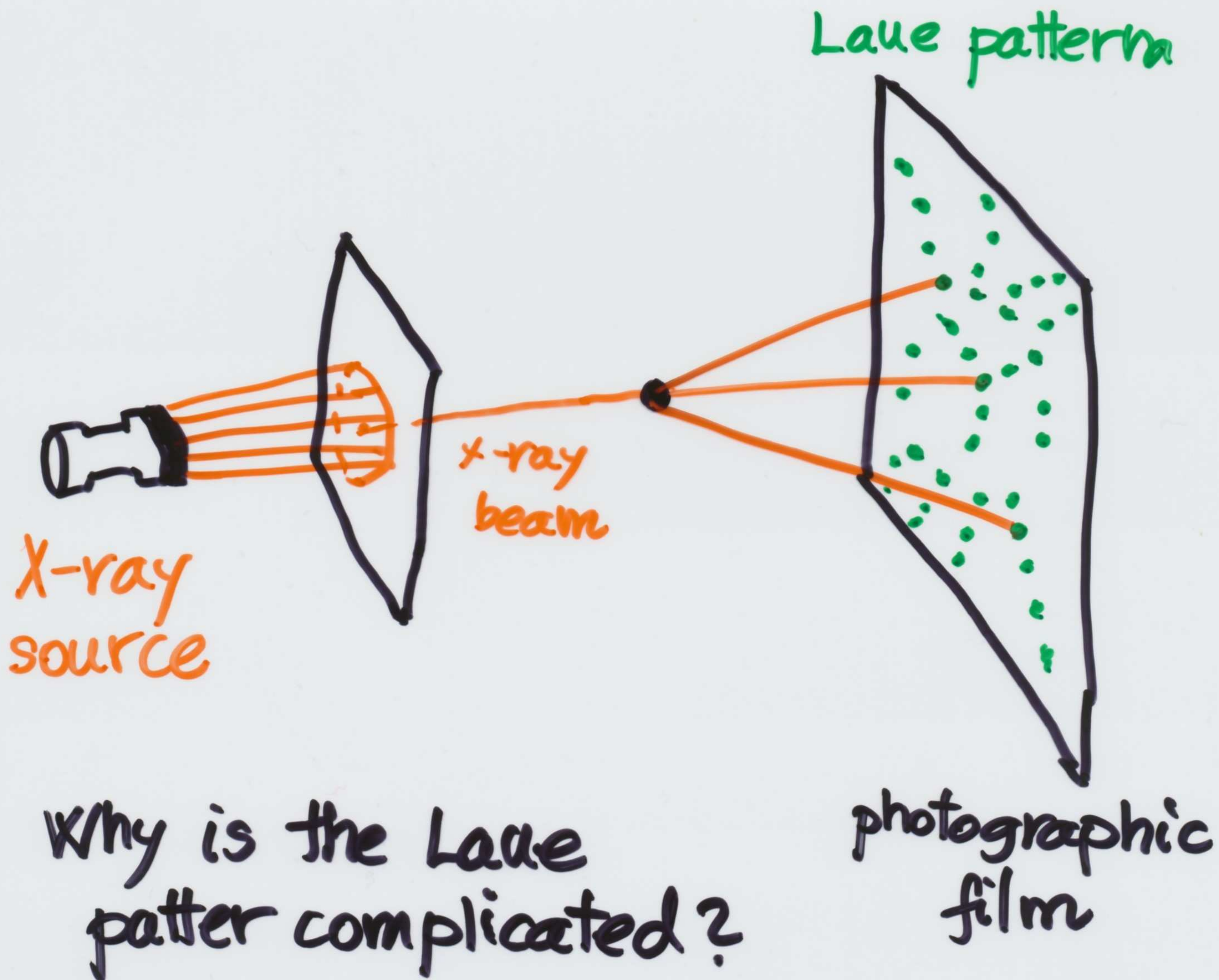
The wavelength λ of any EM wave can be determined if a grating of the spacing $\sim \lambda$ is available

X-rays $\lambda \approx 10^{-10} \text{ m} = 0.1 \text{ nm}$

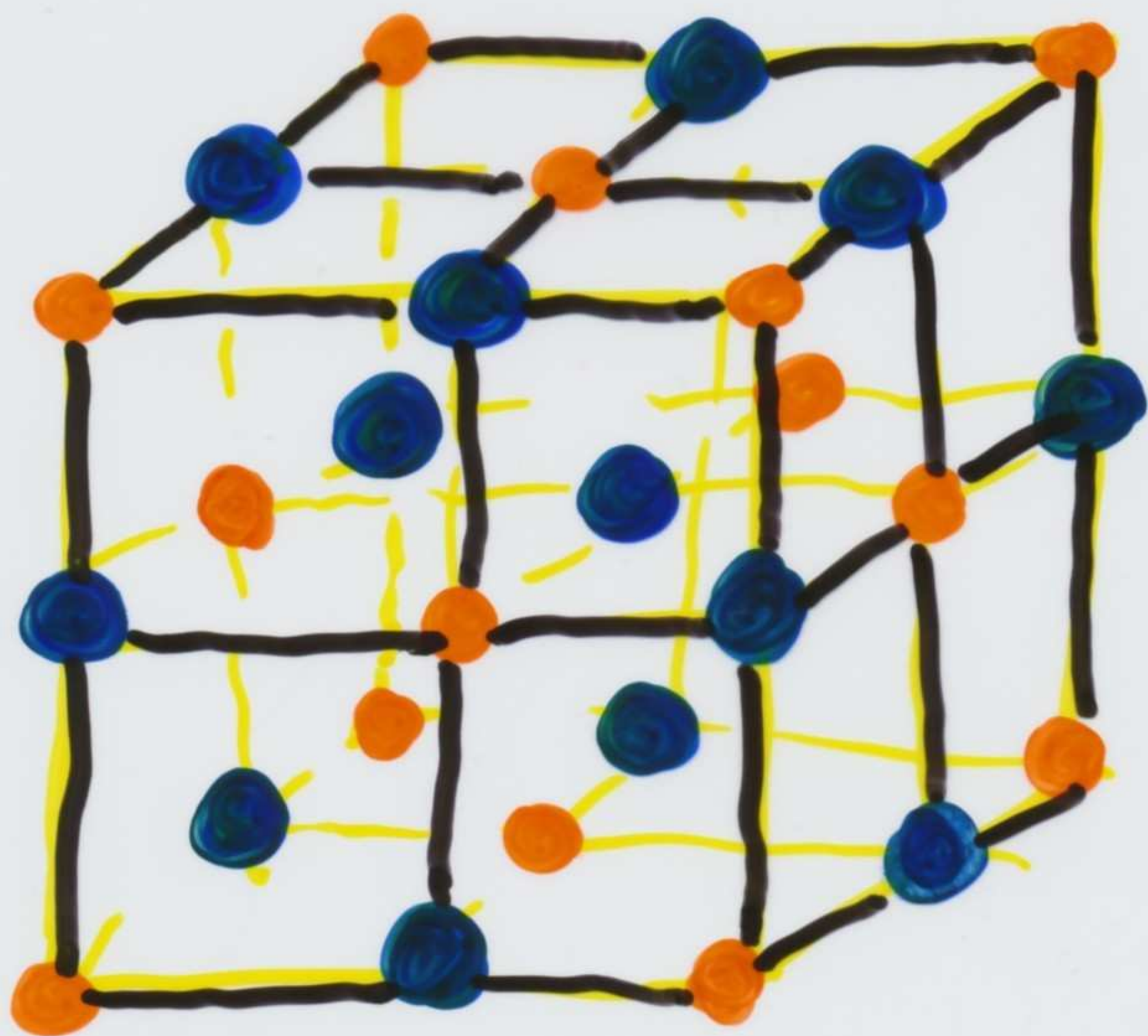
10^{-10} m is the order of inter-atom spacing in a crystal.

Max von Laue suggested that a crystal can act as a 3D grating for X-rays and yield a diffraction pattern.

X-ray diffraction nowadays
⇒ elucidates structure of matter

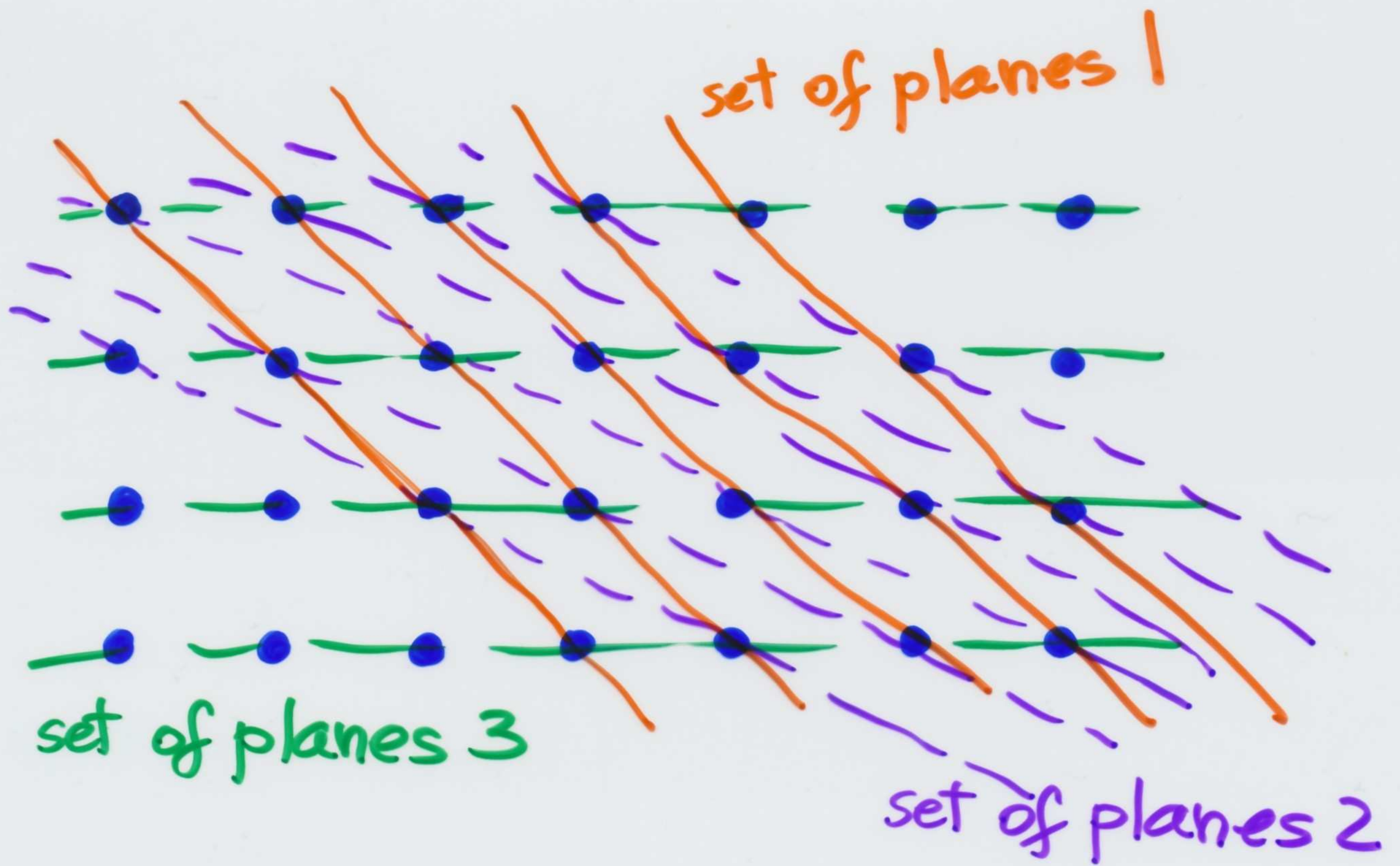


Consider 3D structure of a crystal, for example NaCl:

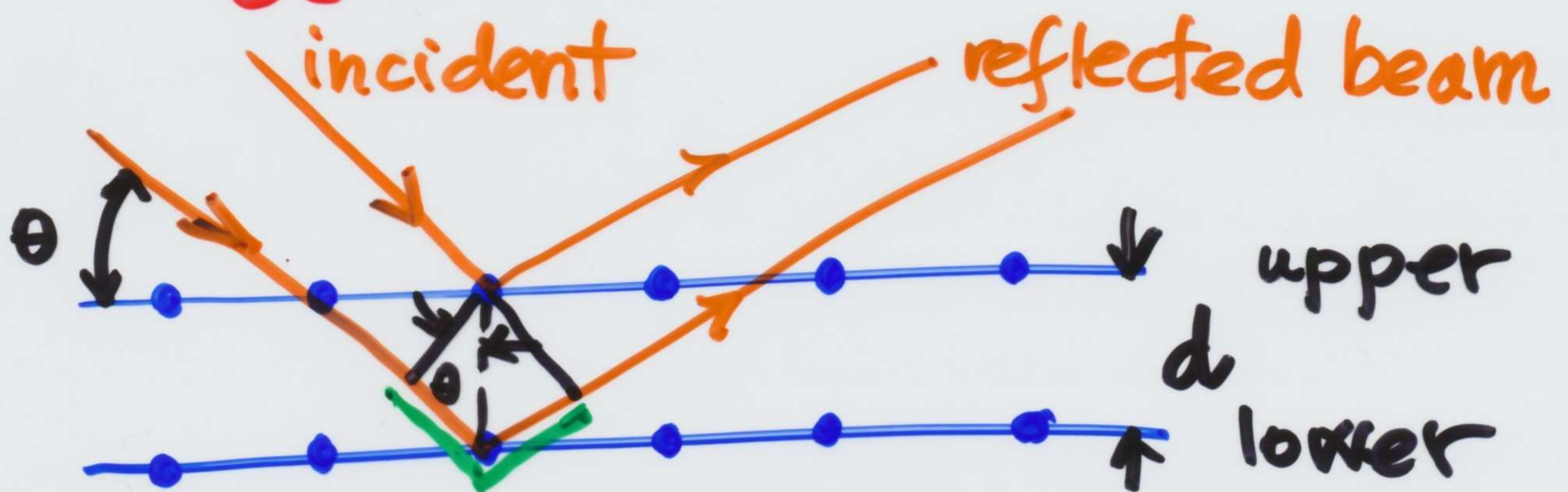


● Na^+

● Cl^-



Bragg's law



The path difference between the upper plane & lower plane reflection:

$$2d \sin \theta = m \lambda$$

$$m = 1, 2, 3, \dots$$

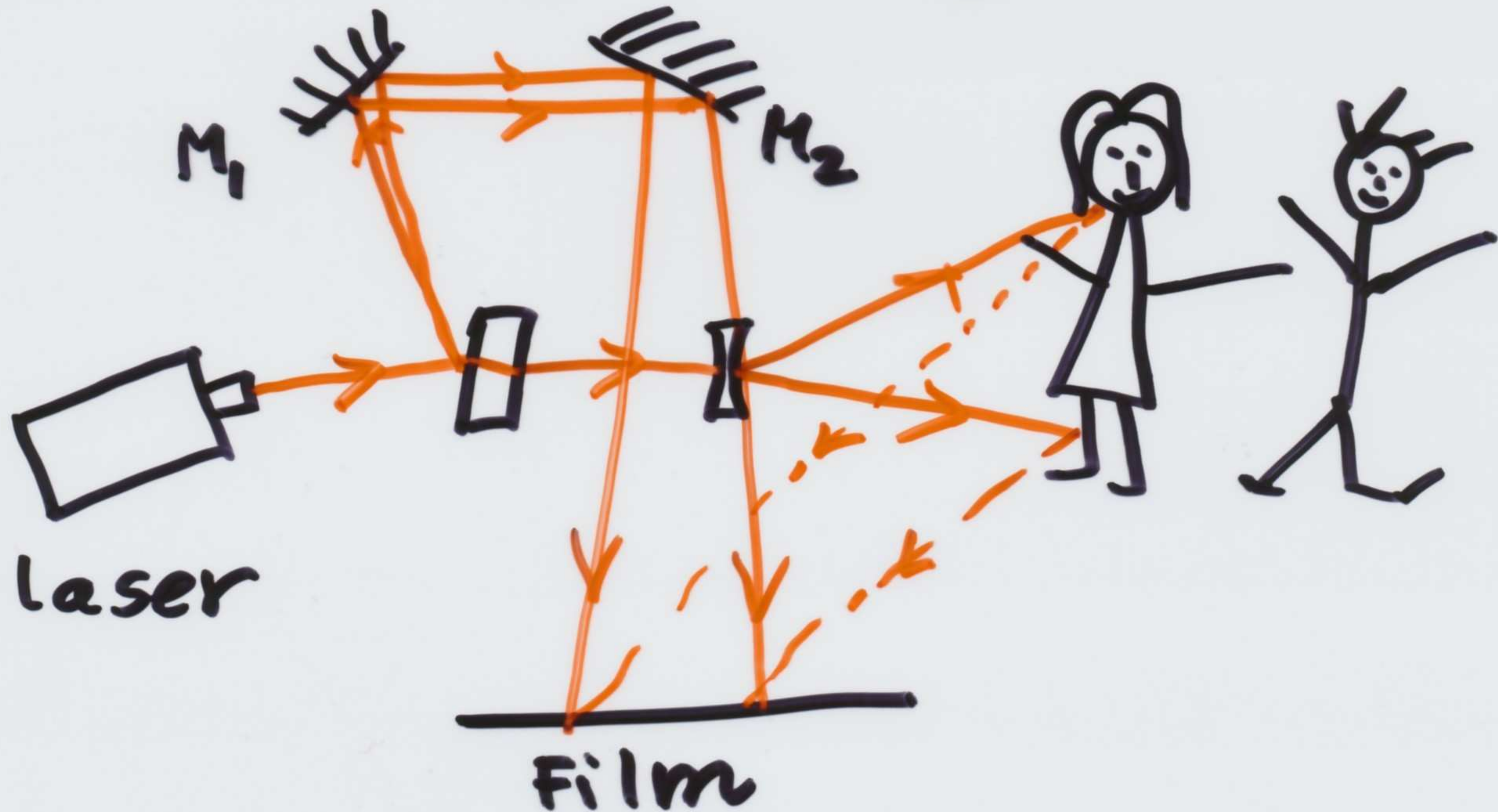
Bragg's law

Holography

Dennis Gabor

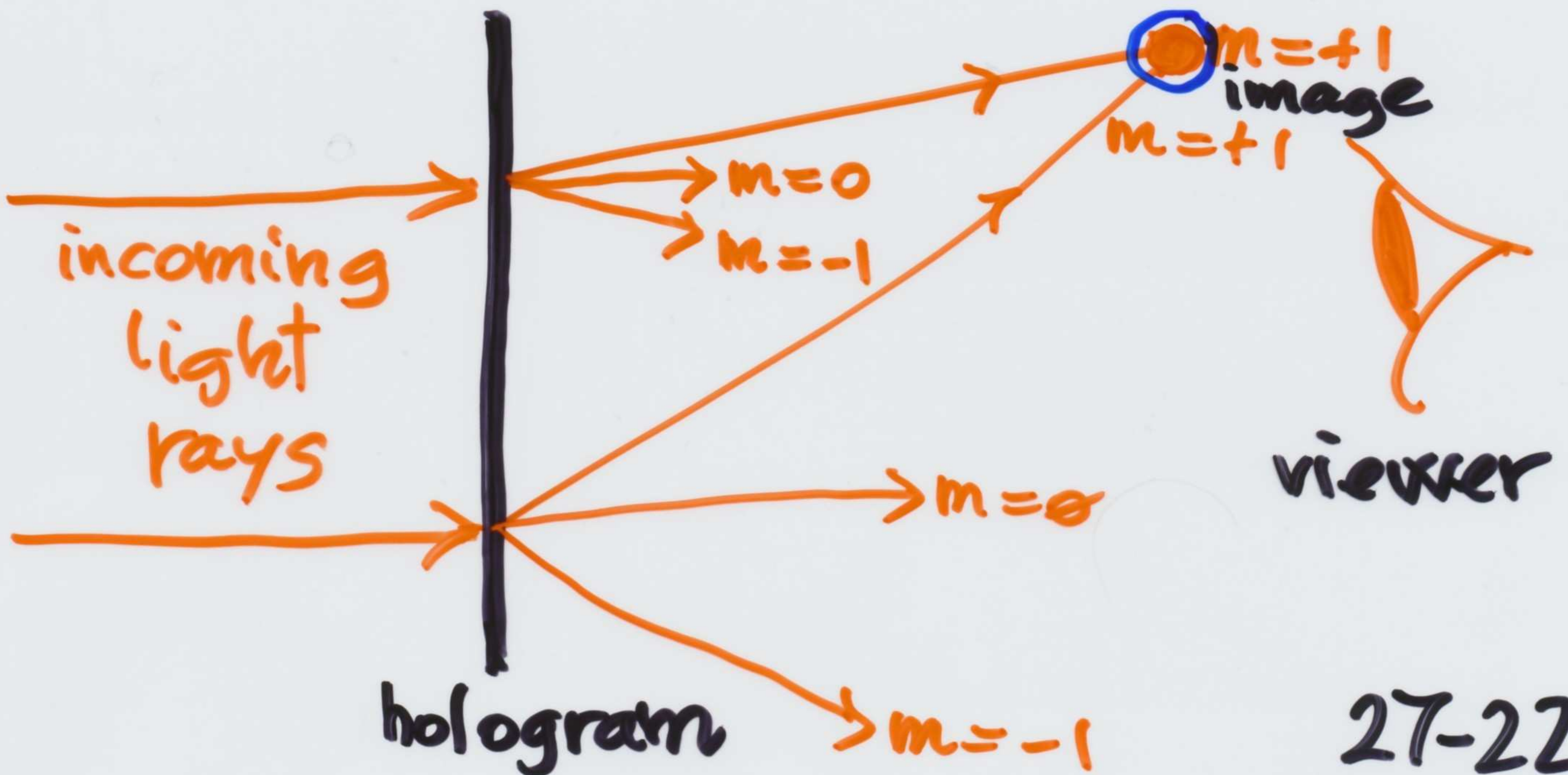
Nobel Prize 1971

(A) Producing a hologram



(B) Viewing a hologram

hologram acts as a grating



Quantum Physics

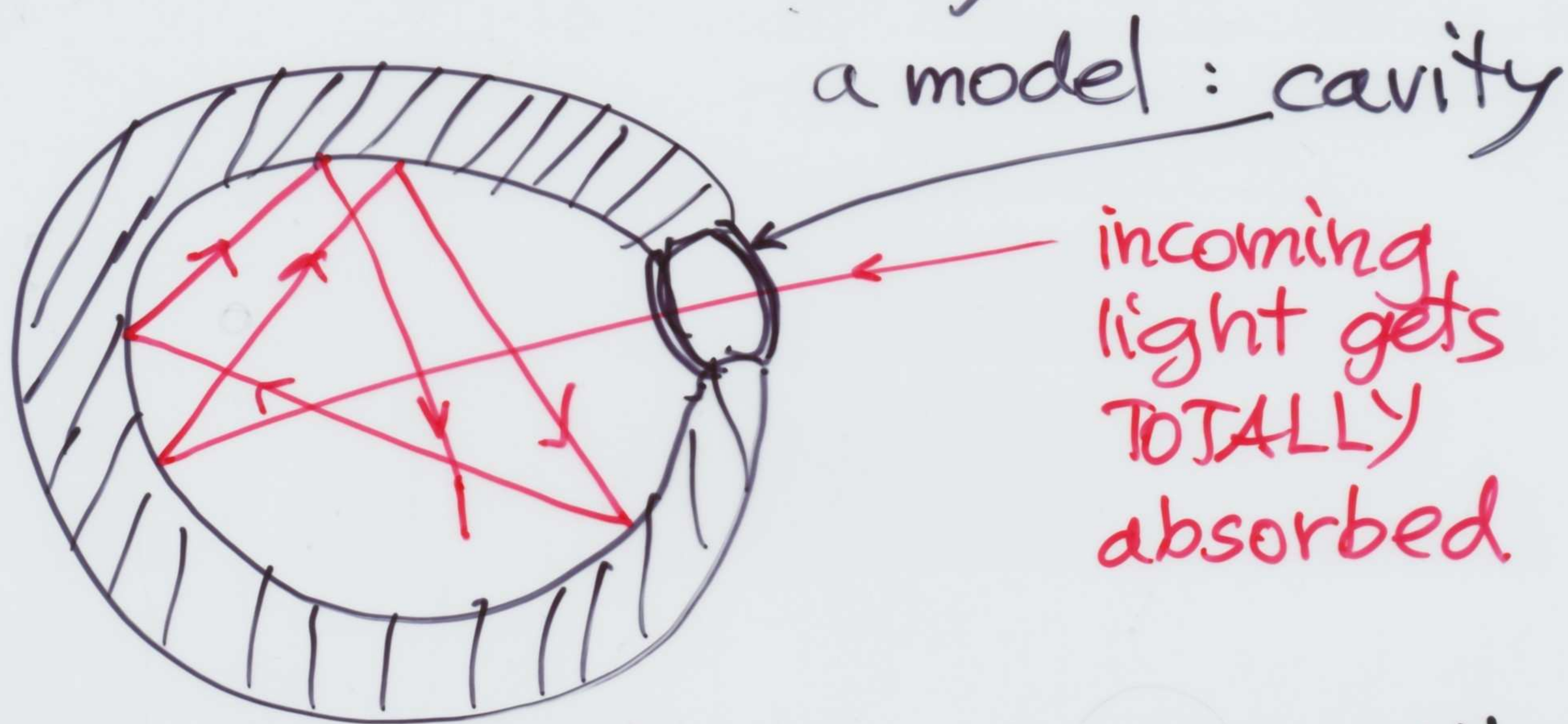
Dual nature of light?

wave vs. particle

Blackbody Radiation & Planck's
Theory

→ any object at temperature $T > 0$
emits energy called **thermal
radiation**

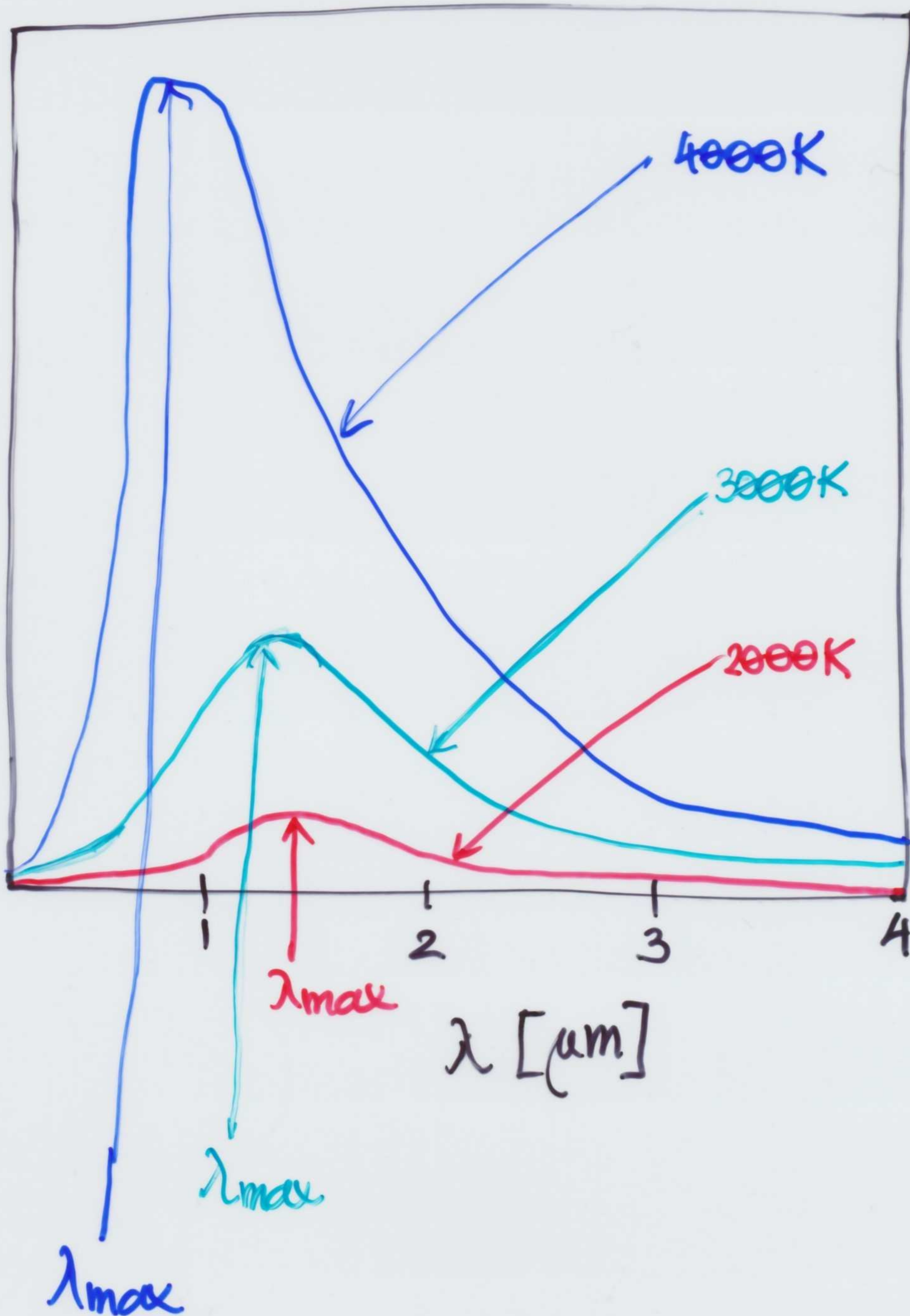
→ what is a black body?



→ the characteristics of radiation emitted
from the hole \Rightarrow temperature of the
walls

→ distribution of different wavelengths of cavity (blackbody) radiation [exp. studied in late 19th century]:

Intensity



Experimentally observed features of the distribution:

→ The total power of emitted radiation increases with temperature ⇒
Stefan's law

$$P = \sigma A e T^4$$

← temperature

← emissivity
(= 1 for blackbody)

← area
(e.g. surface of the black body)

← Stefan's constant $\sigma = 5.67 \times 10^{-8} \text{ W/m}^2$

→ The peak of the wavelength distribution λ_{max} shifts to shorter wavelengths as the temperature increases ⇒
Wien's displacement law

$$\lambda_{\text{max}} T = 2.898 \times 10^{-3} \text{ m} \cdot \text{K}$$

← peak

← temperature

Theoretical explanation for blackbody radiation

→ classically:

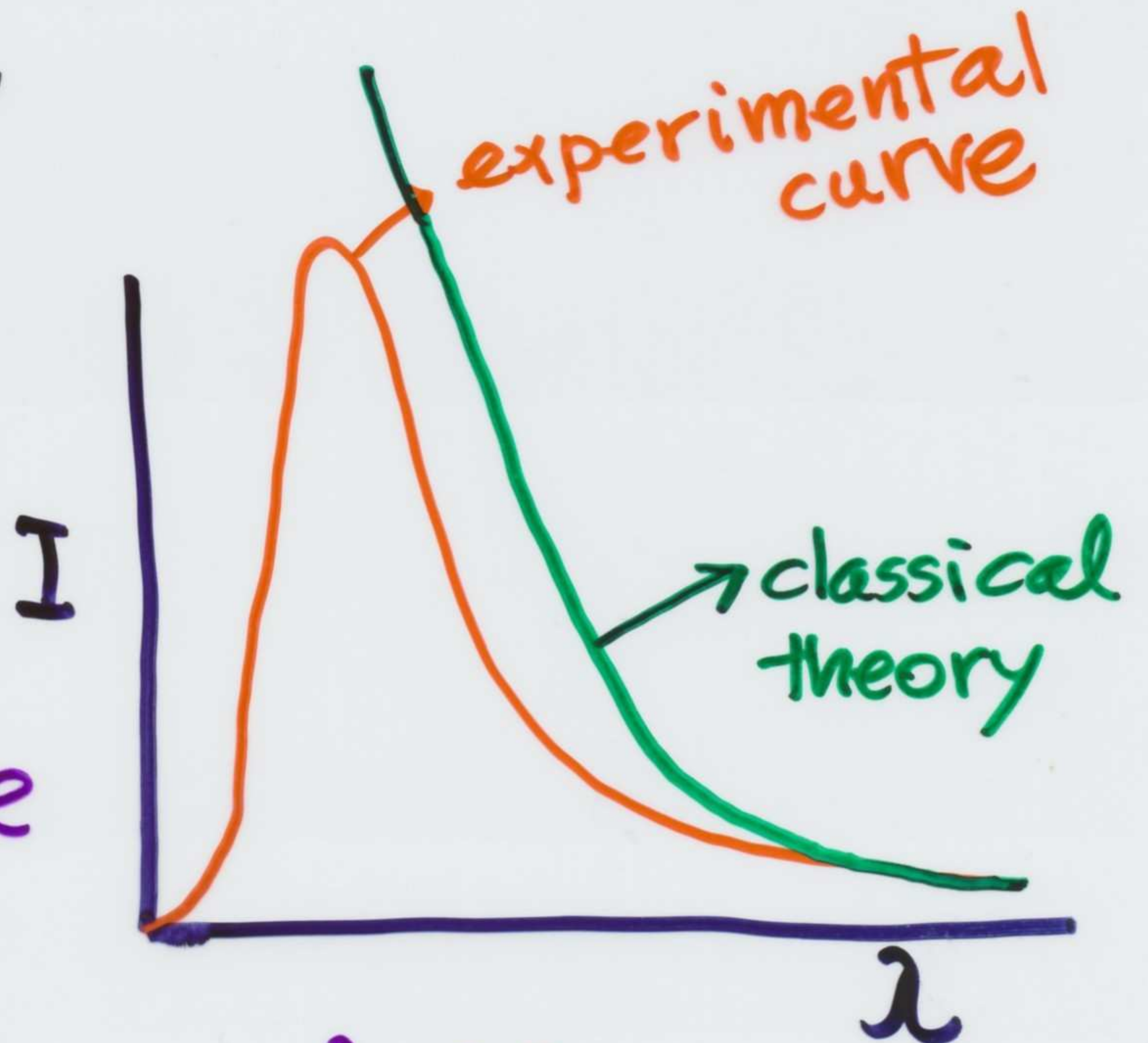
large $\lambda \rightarrow$ agree

small $\lambda \Rightarrow$

ultraviolet catastrophe

small $\lambda \Rightarrow$ UV light

$I \rightarrow \infty$ as $\lambda \rightarrow 0$



→ explanation by Max Planck:

onset of quantum physics

Planck imagined oscillators at the surface of b.b. (related to fluctuating charges in molecules) & made

2 assumptions

(1) → energy of the oscillator is quantized:

$$E_n = n \cdot h \cdot f$$

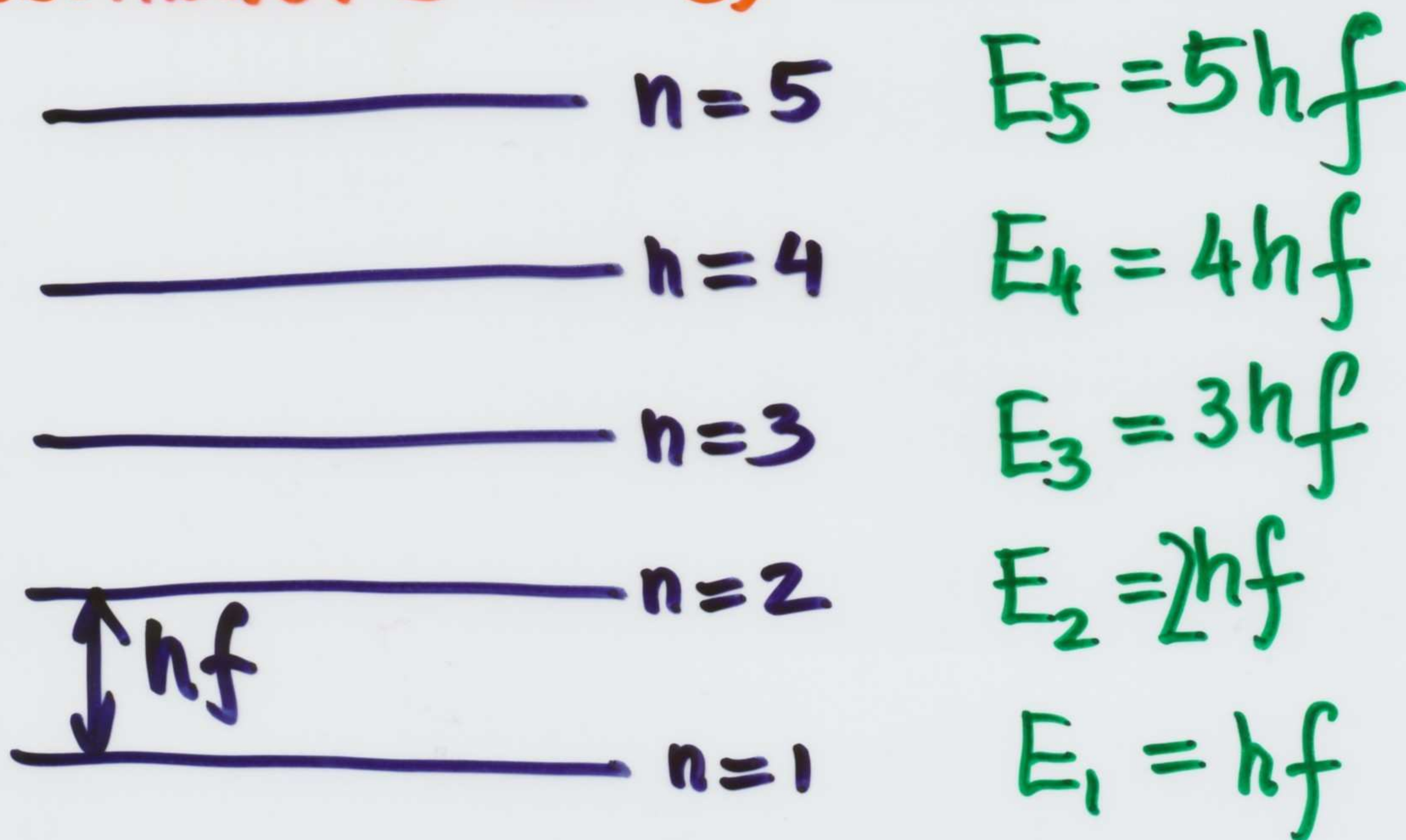
integer

Planck constant

frequency

(2) → oscillators emit or absorb energy in discrete units, quanta of radiation

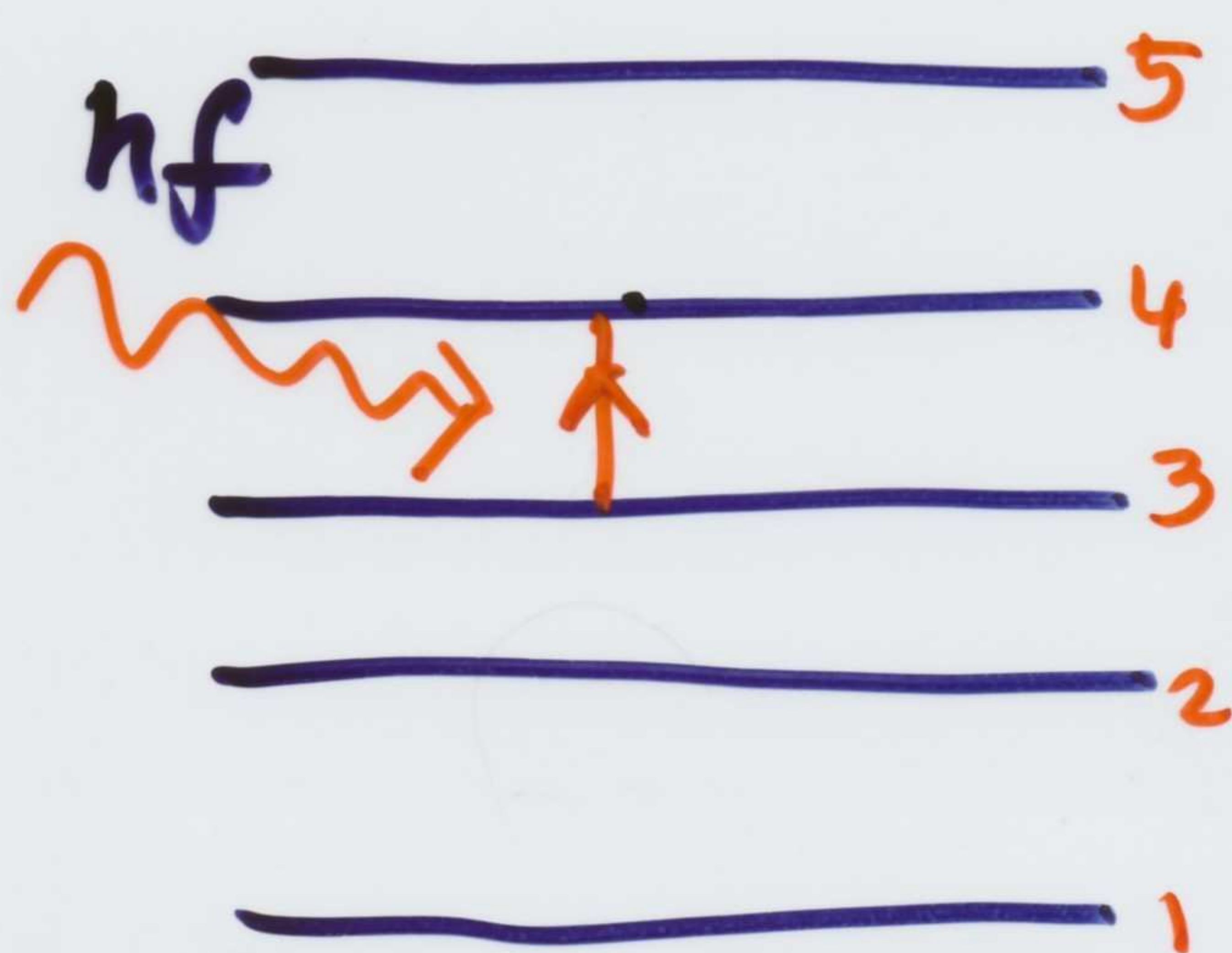
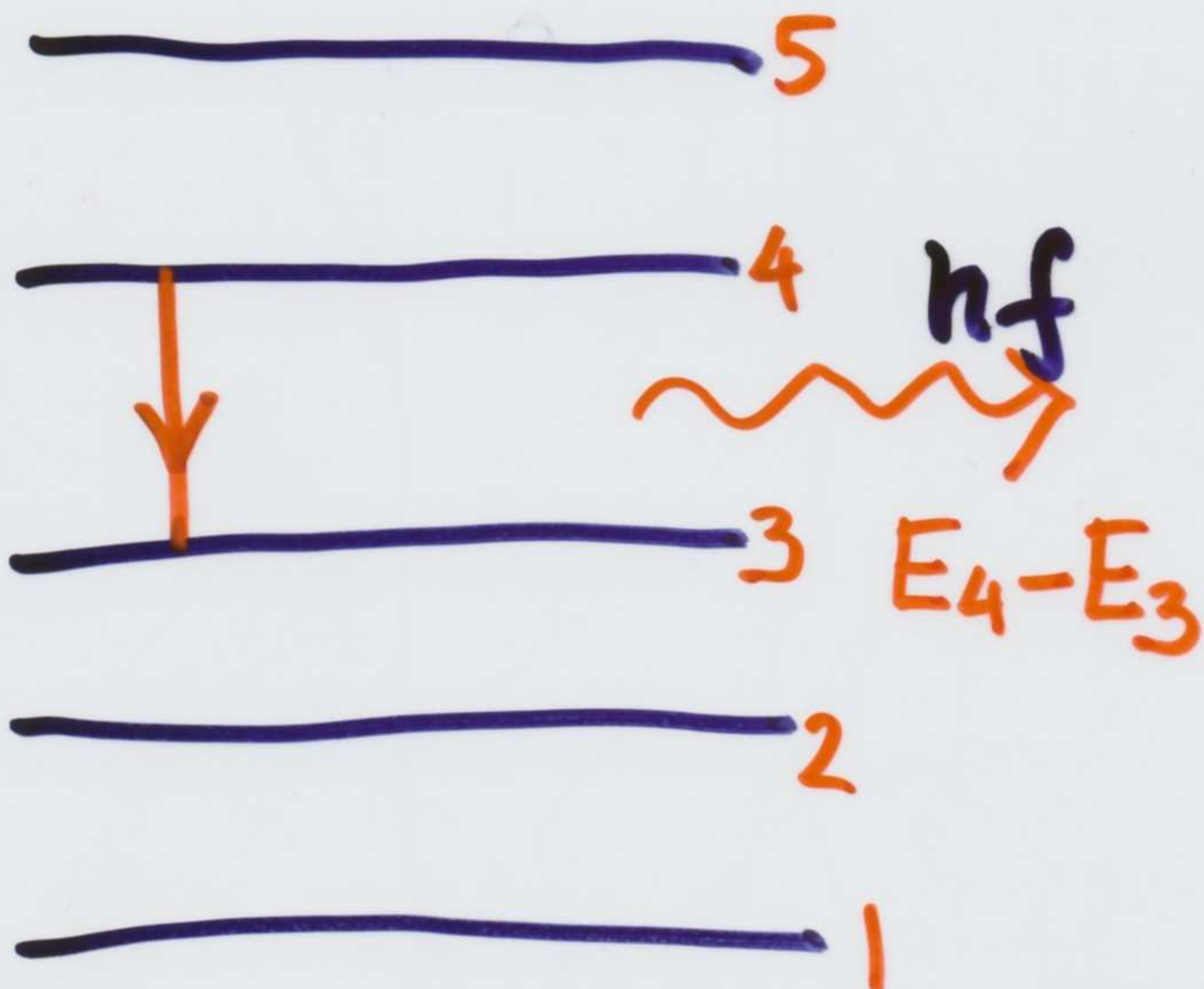
oscillator's energy states : quantum states



quantum transitions

EMISSION $4 \rightarrow 3$

ABSORPTION $3 \rightarrow 4$



Planck presented his theory which yielded the wavelength distribution in remarkable agreement with experimental curves.

Why do we not see quantum effects on a daily basis?

→ in a macroscopic world quantum jumps (say in energy) are so small that our senses perceive continuous behavior

→ quantum effect observed on the submicroscopic level of atoms & molecules

Quiz: In the constellation Orion, stars Rigel & Betelgeuse glow in blue & red light, respectively. Which one has a higher surface temperature?

EXAMPLE: Thermal radiation from the Human Body

The temperature of the surface of the human body is 35°C . Calculate λ_{max} , the peak wavelength, of the radiation emitted by the human body, & estimate the total power \mathcal{P} .

→ Wien's displacement law

$$\lambda_{\text{max}} T = 2.898 \times 10^{-3} \text{ m K}$$

→ $T_{\text{HB}} = 35^{\circ}\text{C} = 308 \text{ K}$ (273 + 35)

$$\lambda_{\text{max}} = \frac{2.898 \times 10^{-3} \text{ m K}}{308 \text{ K}} =$$

$$= 9.41 \times 10^{-6} \text{ m} = \underline{\underline{9.41 \mu\text{m}}}$$

→ infrared region of spectrum

→ Stefan's law: $\mathcal{P} = \sigma A T^4$ } $\Rightarrow \underline{\underline{\mathcal{P} \approx 10^3 \text{ W}}}$

→ area of HB: $A \doteq 2 \text{ m}^2$

height \rightarrow 2m \times width \rightarrow 0.3m \times depth \rightarrow 0.2m