

Lecture 6:

Secondary Structures of Polypeptides

Lecturer:

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Regular Secondary Structures:

- helices (right-handed, left handed)
- β -structures (strands \rightarrow sheet, parallel versus anti-parallel)
- polyproline helices (no hydrogen bonding!)

Irregular Secondary Structures:

- β -turns
- β -bulges

Experimental determination of the secondary structure:

- X-ray spectroscopy
- NMR spectroscopy
- circular dichroism (CD) spectroscopy

Types of helices (R versus L-handed):

→ $i \rightarrow i+2 - 2_7$ helix

→ $i \rightarrow i+3 - 3_{10}$ helix →

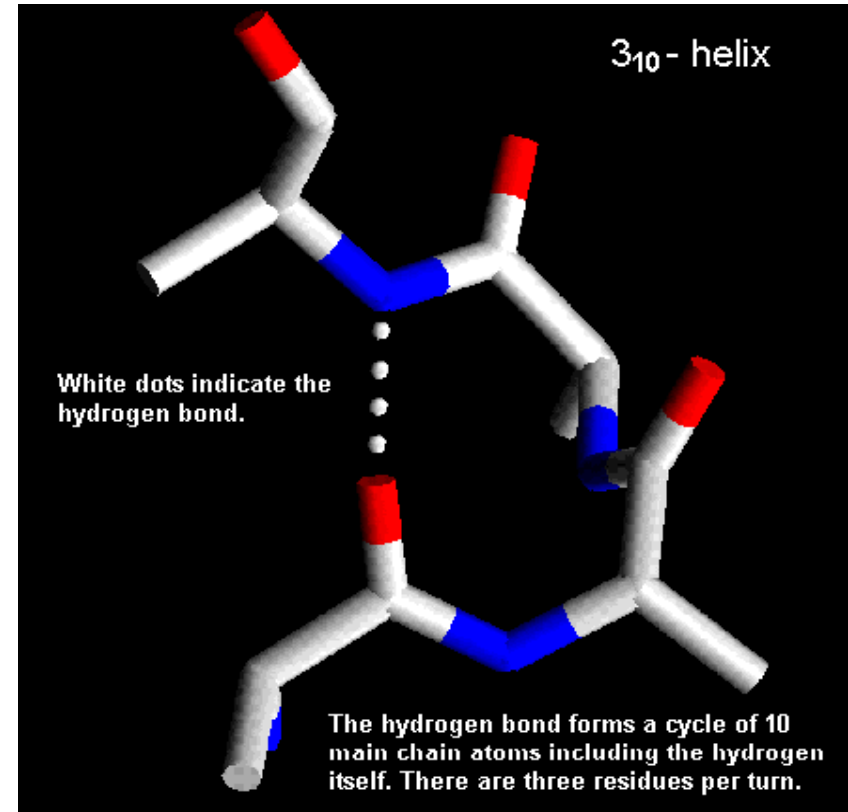
→ $i \rightarrow i+4 - 4_{13}$ helix (α -helix)

→ $i \rightarrow i+5 - 5_{16}$ helix (π -helix)

→ hydrogen bond pattern:

C'O- group of i -th amino acid is H-bonded to the NH-
group of the $(i+n)$ -th amino acid

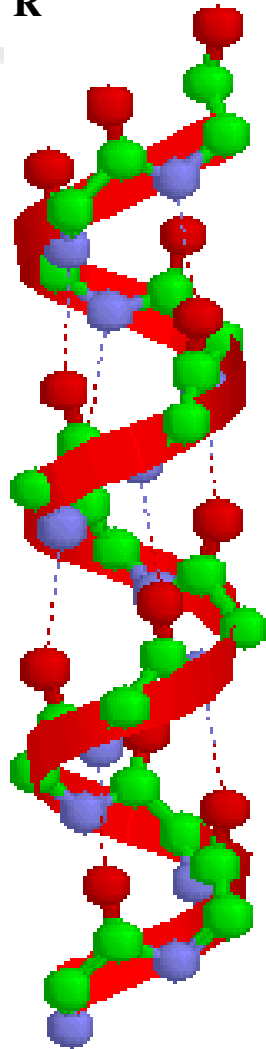
→ the R-handed α -helix = α_R -helix MOST abundant



R-handed \leftrightarrow clock-wise movement \rightarrow away from the observer

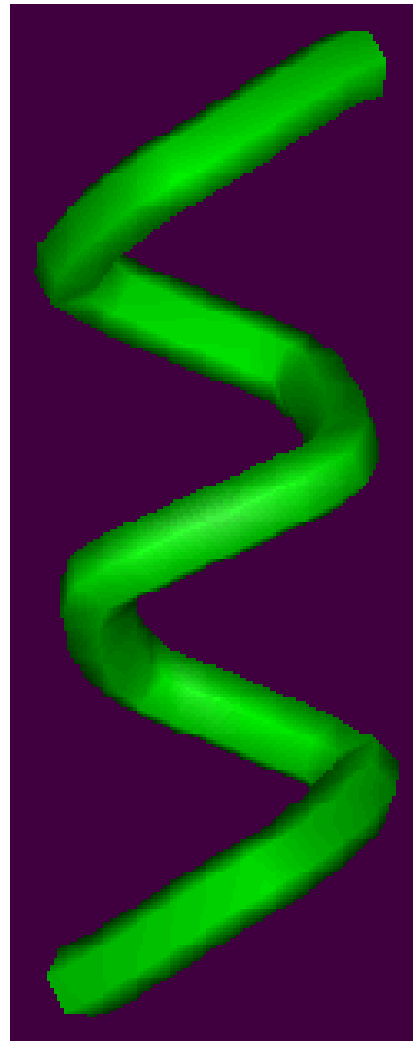
α_R -helix

(a)



5.4 Å

(b)

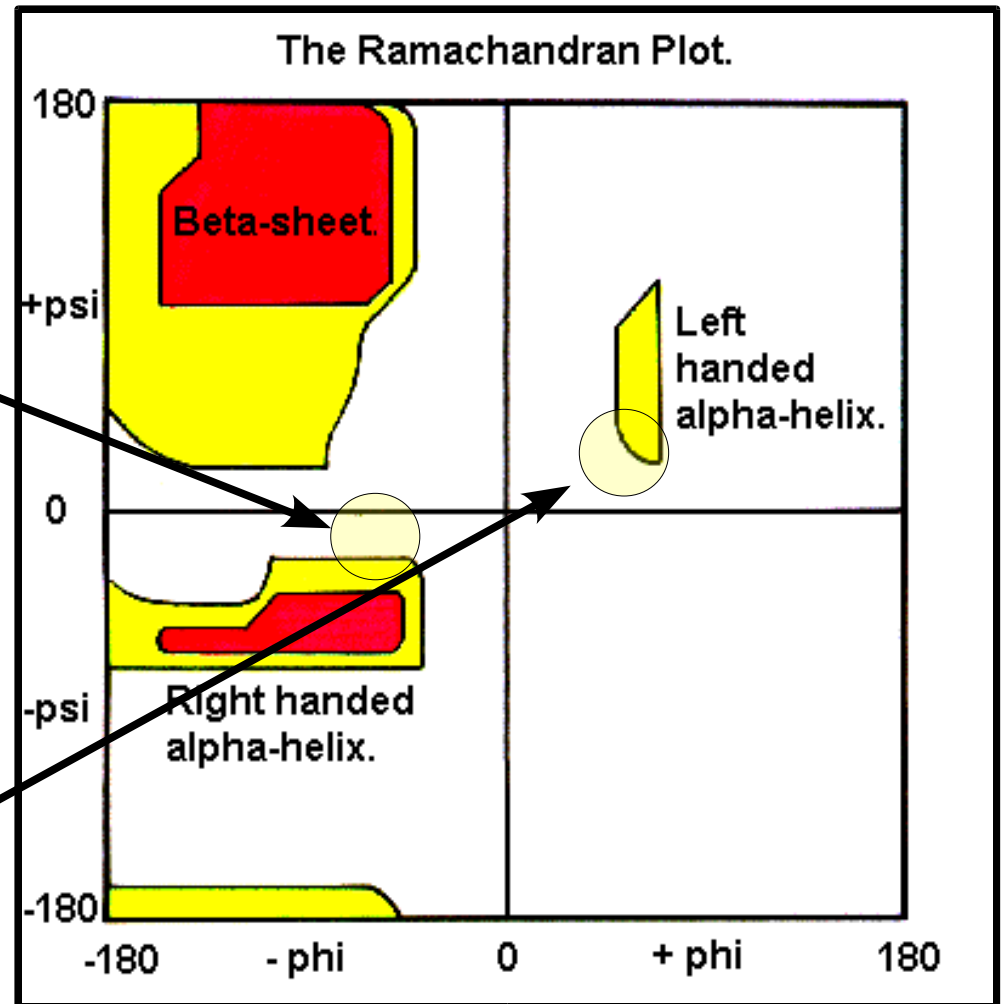


(c)

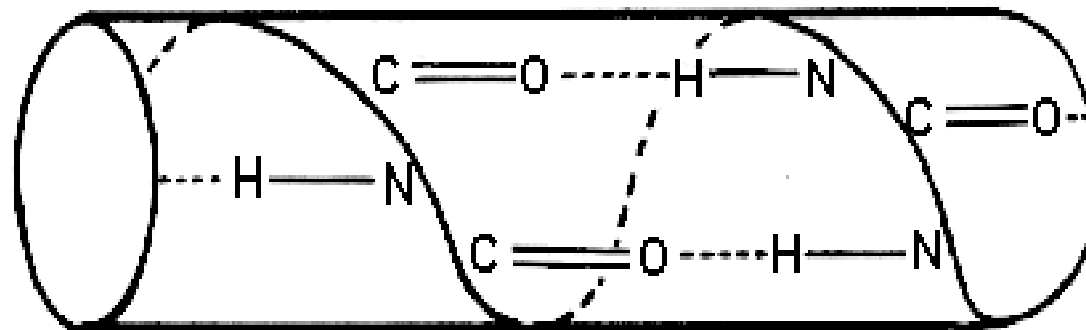


Helix-types present in proteins:

- α_R -helix
- α_L -helix
- short fragments (3-4 aa) of R-handed 3_{10} helix
- L-handed 3_{10} helix (Gly)



Toilet roll representation of the main chain hydrogen bonding in an alpha-helix.



Amino
terminus

Carboxy
terminus

electropositive

N-terminus

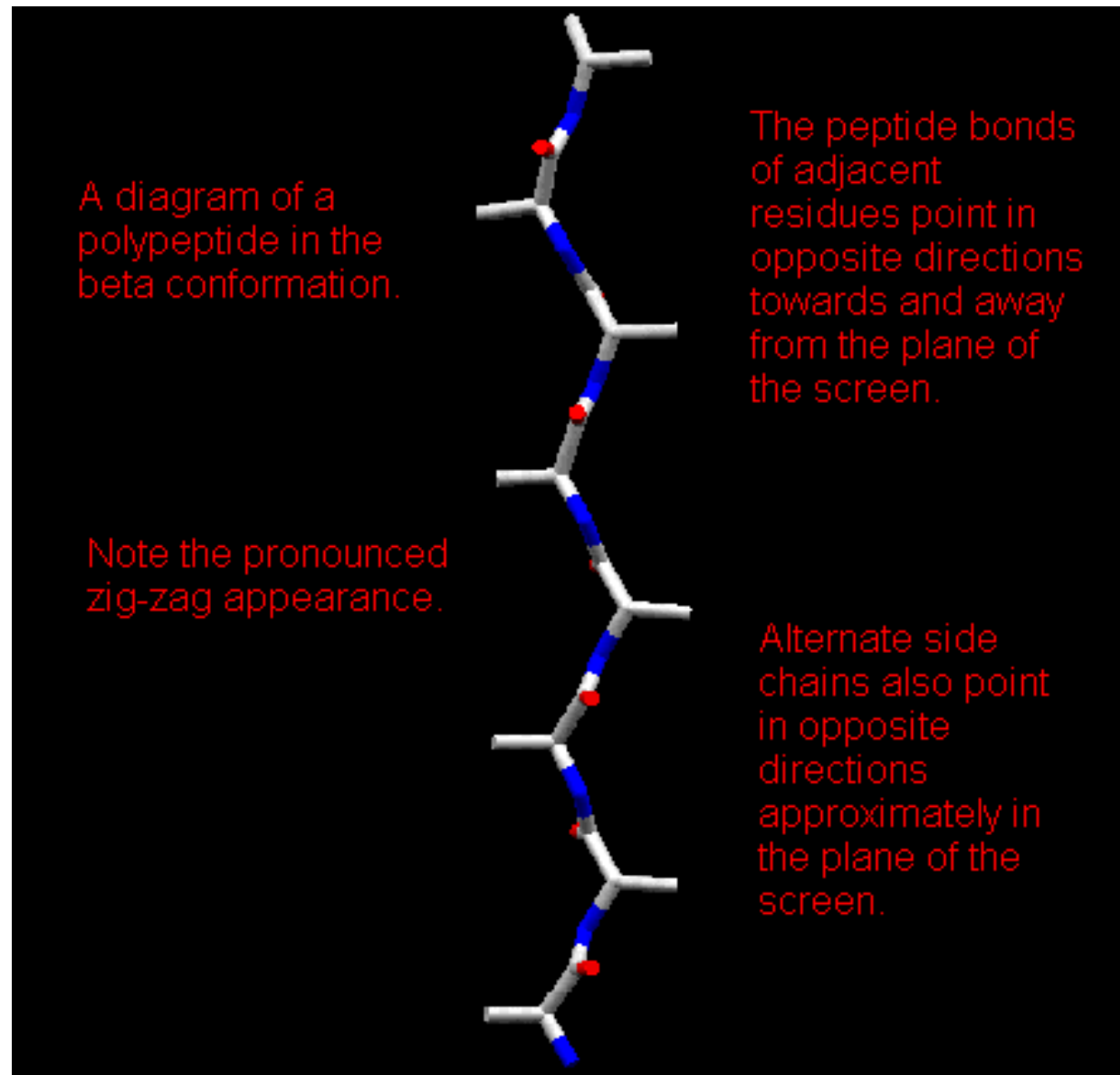
$$+1/2e_0$$

electronegative

C-terminus

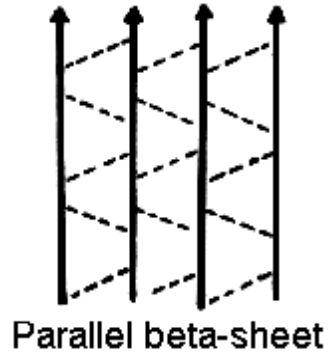
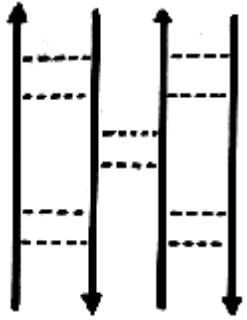
$$-1/2e_0$$

β -strand or extended peptide conformation

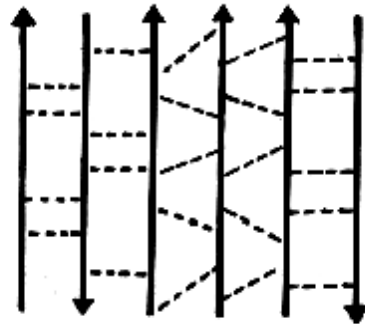


β -Sheet Geometry

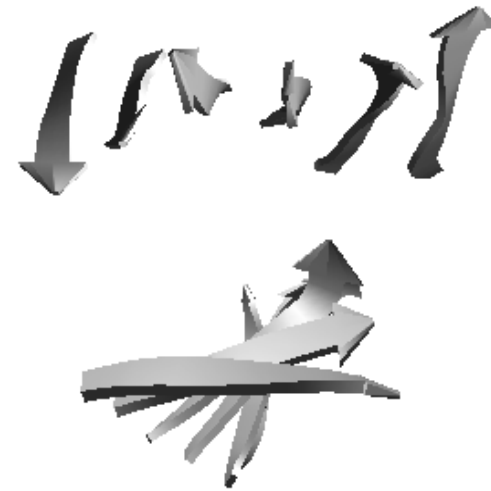
Antiparallel beta-sheet



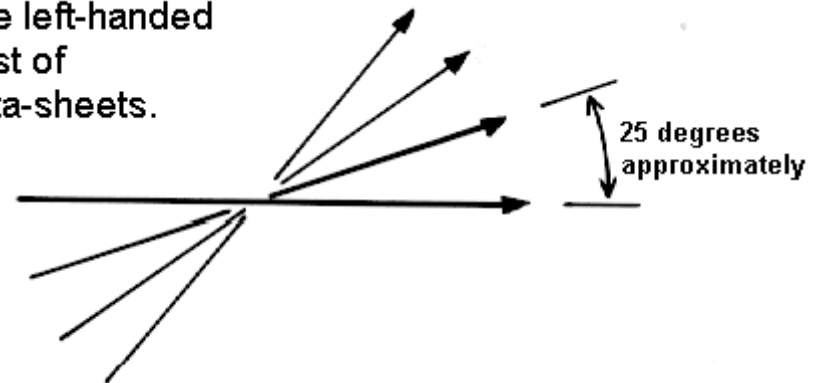
The different types of beta-sheet. Dashed lines indicate main chain hydrogen bonds.



Mixed beta-sheet



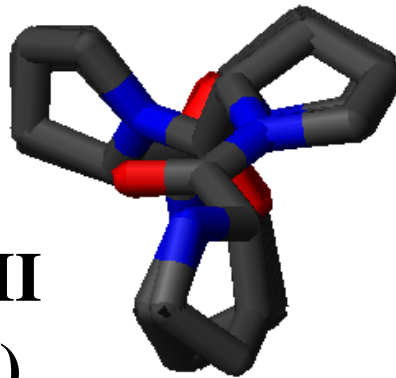
The left-handed twist of beta-sheets.



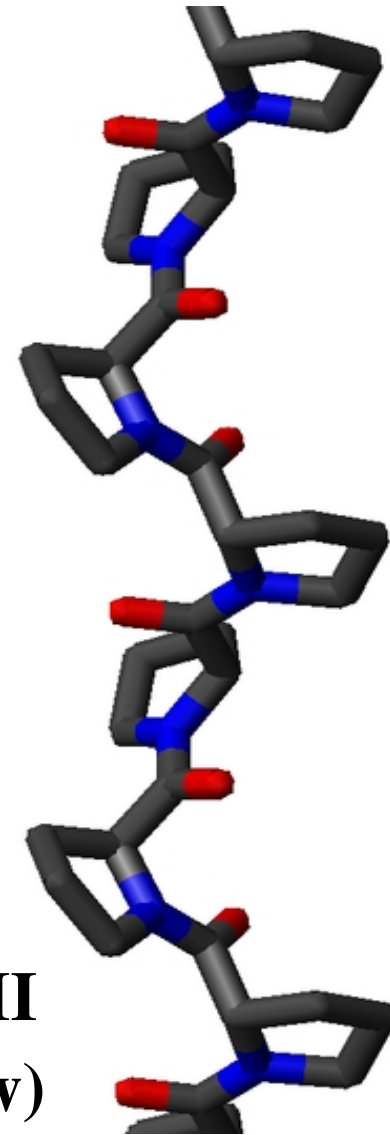
Polyproline helices

- two types, Poly-Pro I and Poly-Pro II
- NO H-bonding
- Poly-Pro II more frequent
- Poly-Pro I includes *cis* conformations (much denser)

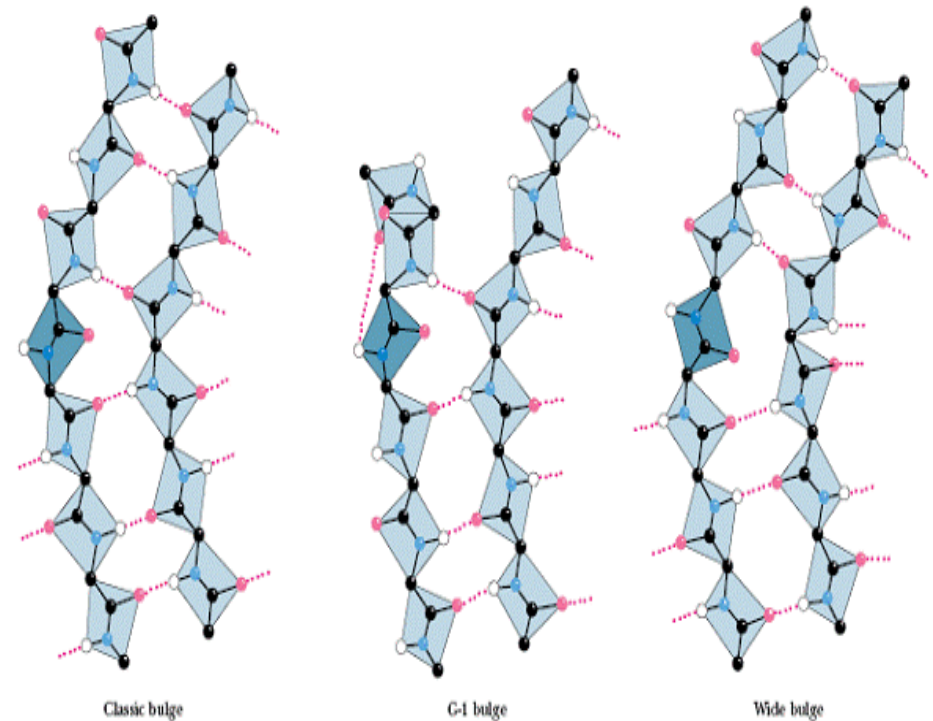
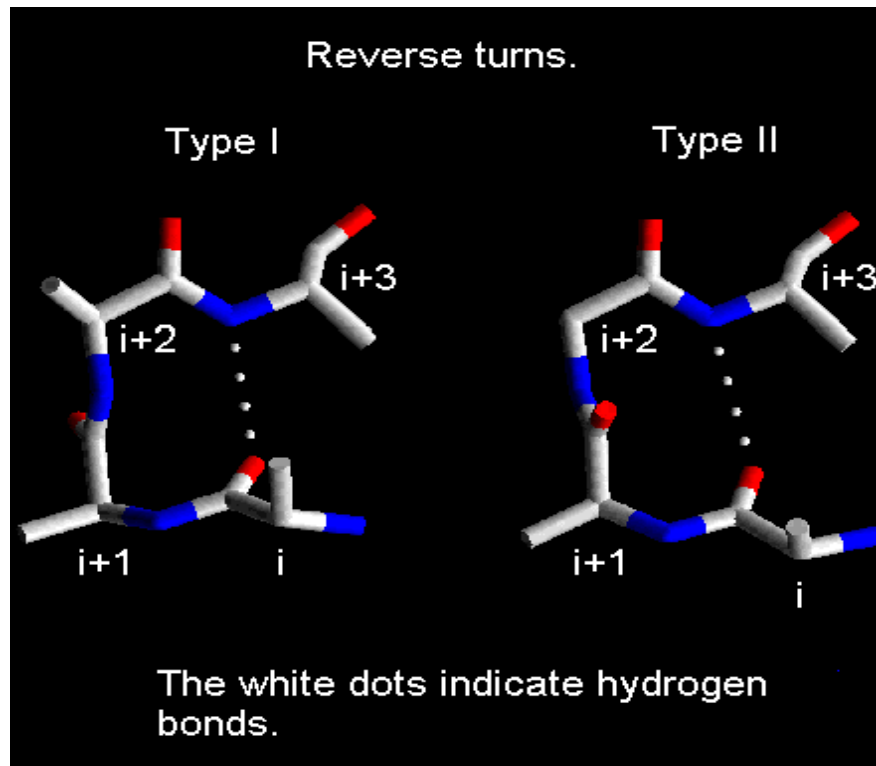
Poly-Pro II
(top view)



Poly-Pro II
(side view)



Irregular Secondary Structures: β -Turns and β -Bulges



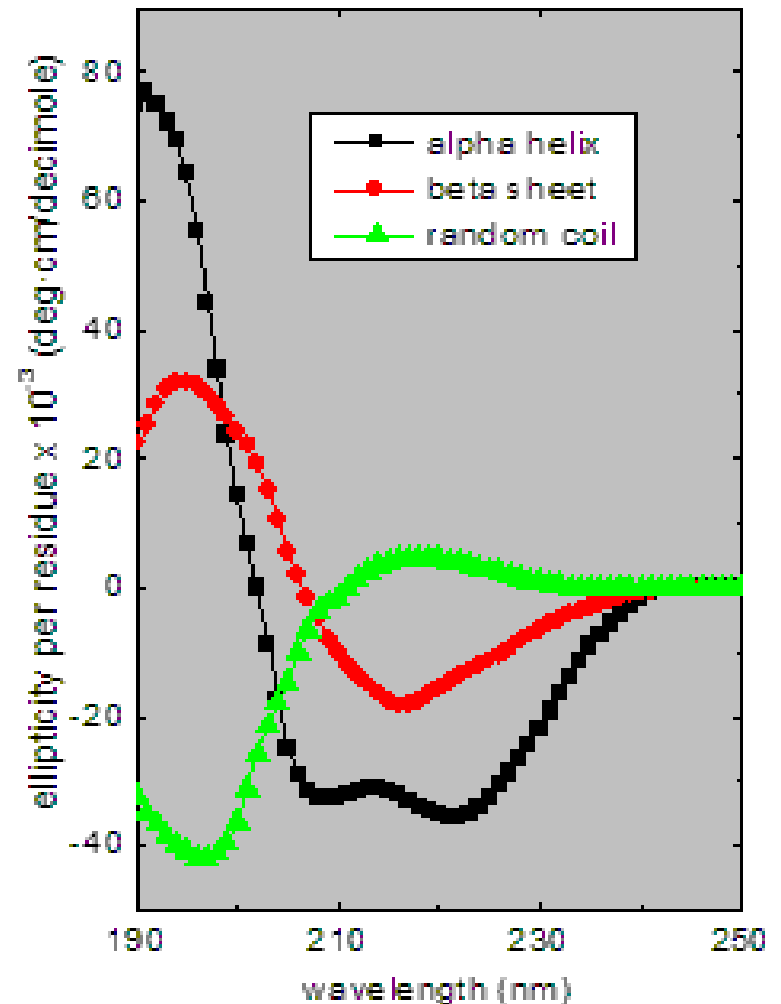
Experimental Determination of the Secondary Structure

Circular dichroism (CD) spectroscopy:

- *no a priori* knowledge on the protein structure needed
- **output: average amounts of α -helical and β -sheet structure**
- **linearly polarized light in \rightarrow elliptically polarized out (clockwise & counter-clockwise polarized light have different absorption, caused by helices of different Handedness)**
- **far UV region (190-240nm): asymmetry in peptide group environment \rightarrow secondary structure**
- **near UV region (250-280nm): asymmetry in aromatic group environment \rightarrow tertiary structure**

Far UV Spectrum for Secondary Structure Determination

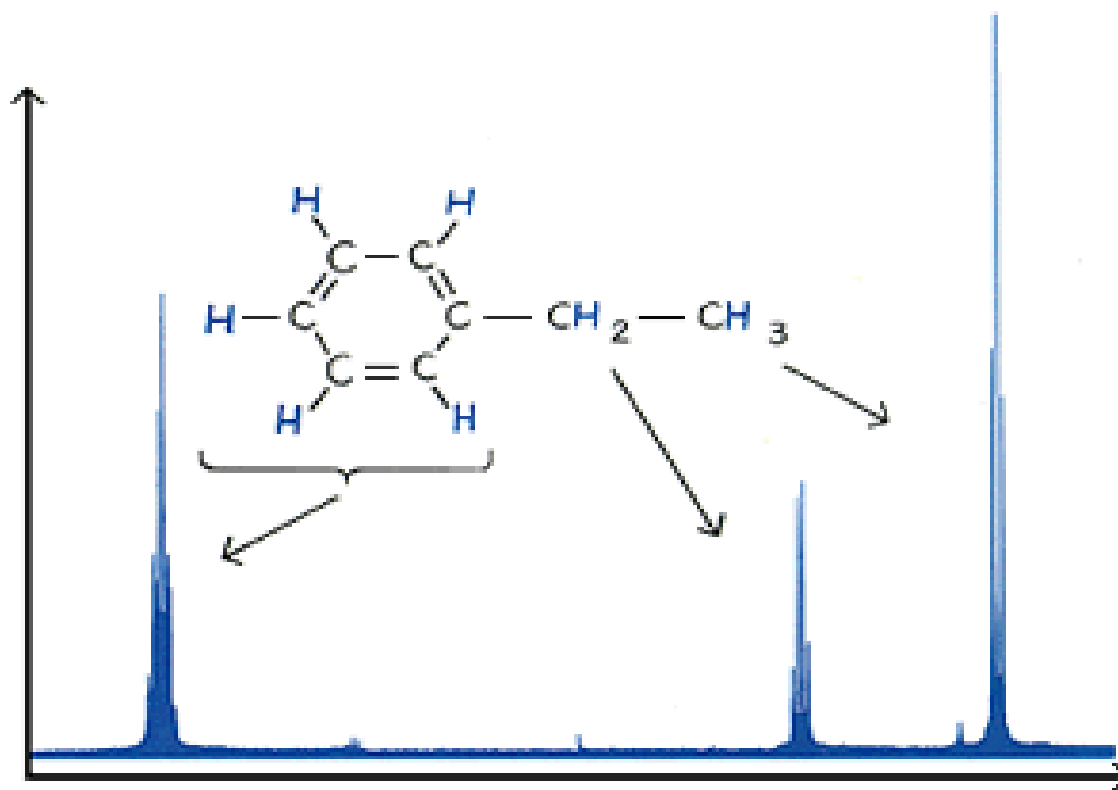
- the chromophore: the peptide bond
- characteristic shapes and magnitudes of the CD spectrum for α -helix, β -sheet, & random coil
- 20–200 μ l of solution with 1 mg/ml to 50 μ g/ml protein, buffer with LOW absorbance in the far UV spectrum
- results in average values of the secondary structures



NMR Spectroscopy

- application of radio—waves to excite magnetic moments of nuclei aligned in a strong magnetic field
- the total number of protons & neutrons need to be odd:
 ^1H , ^{13}C , & ^{15}N
- detects closely positioned H-atom nuclei (4—5 Å)
- resonance at a radio-frequency typical of the nucleus (in a given magnetic field), modified by neighboring chemical bonds
- excitation propagates to the neighboring nucleus giving rise to a cross—peak, which demonstrates proximity of the two nuclei

Example of a simple NMR spectrum:



A proton NMR spectrum of a solution containing a simple organic compound, ethyl benzene. Each group of signals corresponds to protons in a different part of the molecule.